Multivariate statistical characterization of water quality analysis for Erbil wastewater channel

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Abstract: Multivariate statistical analysis, such as principal component analysis (PCA), factor analysis (FA) and cluster analysis (CA), were applied for evaluate the water quality of Erbil wastewater channel. Seventeen variables have been monitored in three selected sites on monthly basis for one year (May 2010- April 2011). Results revealed that the most important variables are those related to salinity factors, temporal hardness and mineral salts according to PCA/FA techniques. In addition to, CA results showed that dissimilarity of site 3 from other two sites, and most similarity between ions related to processes of hydrogeochemical and mineralization.

Generally, water quality for irrigation purposes comes under good water type (86% of total samples) to moderate type (16%) depending on EC- SAR values guidelines. Increases in ions concentrations for EC, TDS and hardness were observed from upstream toward downstream of Erbil wastewater channel.

Keywords: Water quality, factor analysis, principal component analysis, cluster analysis, irrigation, wastewater.

I. Introduction

The safety of human public health and the environment is very important issue related to monitoring of parameters used in assessing water and wastewater [1]. Physico- chemical properties and heavy metals determination for stream water monitoring is necessary to evaluate contamination level in wastewater, with increasing overpopulation, urbanization and different anthropogenic activities, there has been a rapid increase in domestic and industrial effluent discharge into streams and rivers, leading to increase pollution load [2].

Erbil wastewater channel is important stream that recipient of Erbil city untreated sewage, storm runoff rain water, in addition to industrial and agricultural runoff. In medial and downstream it being used for farm irrigation before discharged to Greater Zab River. According to Erbil city general directorate of water in 2012 they produced about 61000 m³.day⁻¹ and every person in city will receive more than 400 litter.day⁻¹ for domestic consumption. Therefore, the quantity of sewage discharge differs from time to other, as a mean, it may reach 77760 m³.day⁻¹ during dry season and 108000 m³.day⁻¹ during rainfall season [3].

The application of different multivariate statistical techniques, such as cluster analysis (CA), principal component analysis (PCA) and factor analysis (FA) helps in the interpretation of complex data matrices to better understand the water quality and ecological status of the studied systems, allows the identification of possible factors sources that influences water systems and offers a valuable tool for reliable management of water resources, as well as rapid solution to pollution problems [4].

Many studies have been conducted on Erbil wastewater monitoring and its suitability for different purposes [5; 6; 7; 8; 9; 10; 11; 12].

The objects of present work are to assess levels of physico- chemical parameters in order to determine the source of pollutants and on quality of water in stream by using multivariate statistical methods, as well as, to evaluate their suitability for irrigation purposes.

2.1 Study area

II. Materials and methods

Erbil wastewater channel which extended from southwest of Erbil city with their elongation for more than 50 km passes through vast farmlands, orchards and several villages, after Gameshtapa village wastewater effluent discharges into Greater Zab River. Generally, Erbil sewer system is constructed for storm water and in most cases domestic sewers are connected illegally with storm sewer. The width of channel ranges from 2-4m with depth more than 1m at different locations [12].

2.2 Sample collection

In present investigation, water samples were collected regularly on monthly basis from 3 sites along Erbil wastewater channel during May 2010- April 2011 (Fig. 1). Standard techniques were used [13] to analyze the different physico- chemical parameters: pH, electrical conductivity in field by using (pH meter Philips 4014 and EC meter Philips 4025 respectively), total dissolved solids (TDS, evaporation at 180 °C), total hardness,

calcium and magnesium (EDTA titrimetric method), Cl (argentometric method), SO_4^{-2} (turbidimetric method), HCO₃ (titrimetric method), Na⁺ and K⁺ (flame photometric method), trace metals (Cd, Pb, Cu and Zn) by (Atomic absorption spectroscopy Philips Sp9) after concentrating 1L of samples (with addition of HNO₃). While, sodicity, including the Na% and SAR were calculated according to the formulas given by [14] as follow:

Sodium percentage % Na =
$$\frac{\text{Na x 100}}{\text{Na + K + Mg + Ca}}$$

Sodium Adsorption Ratio (SAR) = $\frac{1}{\text{SAR}} = \frac{1}{\text{SAR}}$

$$aR = \frac{Na}{\sqrt{\frac{Ca + Mg}{2}}}$$

2.3 Multivariate statistical methods

Physico- chemical factors were compared statistically by using one way analysis of variance (ANOVA). Duncan test was used to examine multiple comparisons. Correlation was evaluated for significance using a Pearson product moment correlation test procedure. Principal component analysis/ factor analysis (PCA) was performed to relate different studied parameters and cluster analysis (CA) to evaluate similarities and dissimilarities between sampling sites and variables, all mathematical and statistical calculation were implemented using (SPSS version 20) program and Microsoft Office Excell 2007.

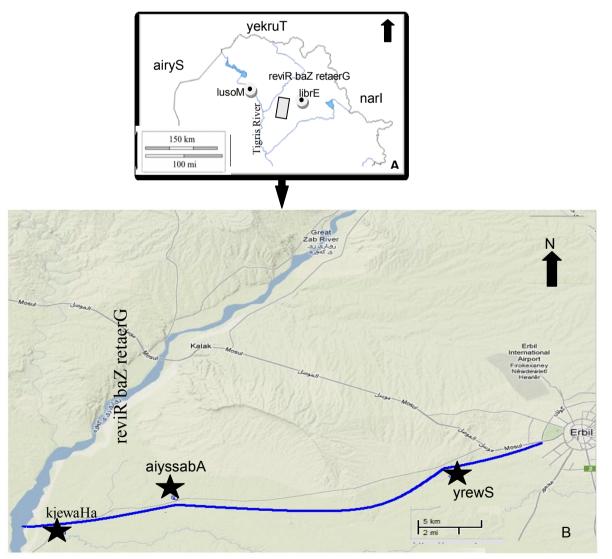


Figure 1: Map showing northern part of Iraq, and sampling sites along the Erbil wastewater channel.

Descriptive characteristics (minimum, maximum, mean and standard error) for chemical parameters are given in Table (1). The sewage water samples exhibited an alkaline pH in the range of 6.99 to 7.98. The minimum value observed at site 1 and maximum value at site 3 (Fig. 2). There is a significant difference ($P \le 0.05$) between site 1 and sites 2 and 3. Generally, the obtained pH values fall within safe limit for crop production [15]. Previous studies [5; 9; 16] recorded similar results in the main sewage channel of Erbil city.

Conductivity is a measure of the ability of water to conduct an electrical current [17]. The electrical conductivity (EC) of water samples ranged from 486- 1032 μ s. cm⁻¹ throughout the studied period. Higher EC were observed in downstream channel (site 3) with trend to increase from site 1 to site 3, it may be related to effluent discharges into the stream [18]. Based on irrigation purposes in most cases EC classified as C2 medium type water and some cases as C3 saline type water.

Sodium Adsorption Ratio (SAR) is a relatively proportion of sodium ions to calcium and magnesium. Sodium effects soil structure through cation absorption, mainly on clay and fine silt fractions. As the percentage of sodium rises in relation to other cations, lead to reduced infiltration that cause a decrease in water supply to crop between irrigations and reduced soil aeration capacity [15]. SAR values ranged from (0.32- 1.84, 1.14- 2.16 and 1.14- 1.80 Meq.l⁻¹) for sites 1, 2 and 3 respectively. The water quality characterized by low SAR and lies in first class (S1) according to [19] classification, which can be considered as safe type for irrigation, since the maximum SAR value not exceed 3. These results are lower than obtained by [20]. Generally, about 84% of the total water samples fall in the C2- S1 zone of the salinity that regarded as medium salinity type. This comes under the good water quality zone for irrigation. While, 16% of water samples fall in the C3- S1 (high salinity type) that comes under moderate water quality zone for irrigation (Fig. 3) [19].

Total dissolved solids (TDS) showed same trend of fluctuation as noticed in EC values by increases from upstream to downstream of sewage channel (Fig. 4). TDS mean values were 345, 380 and 387 mg.l⁻¹ respectively. The relative high ranges of EC and TDS values in the sewage water may be related to nature of municipality pollutants, industrial wastes and land use activities in the area [21]. High TDS concentrations automatically influenced the quality of the received water body. Elevated TDS can be toxic to freshwater animals by causing osmotic stress and affecting the osmoregulatory capability of the organisms [1]. Similar results were noticed by [11; 16], while, its lower than recorded by [8] at the same polluted channel. TDS values it considered as good type and entirely safe for irrigation as classified by [22]. Erbil wastewater can be regarded as a weak type according to [14]. There is high correlation coefficient observed between EC and TDS in this investigation. The TDS is related to conductivity in that if they are high, the EC also be high [23; 24].

Total hardness mean values varies from 277.9, 287 and 296.3 mg CaCO₃.1⁻¹ from upstream to downstream respectively. There is no significant difference (P \leq 0.05) between studied sites. Water quality can be regarded as hard to very hard water [22].

Sodium and potassium are common felsic cation occurrences in nature, with high dissolvent (soluble) in water [25]. The mean values of Na⁺ and K⁺ from upstream to downstream were (1.73, 2.07 and 1.91 for Na and 0.21, 0.22 and 0.20 Meq.l⁻¹ respectively). Highest Na⁺ concentration than K in water samples may be due to domestic effluent characteristic that contain higher concentration of Na⁺ [26] or to higher resistance of K minerals to weathering in relation to Na minerals [27]. There is a trend to increase concentration for both ions from site 1 to site 2 then reduction toward site 3. Same phenomenon was observed by [5] at same polluted channel. Higher values of Na⁺ and K⁺ was recorded by [8; 28]. While, similar ranges were observed by many researches [5; 9; 10] in the same polluted channel.

Calcium regarded as an important and dominant cation in water, magnesium comes after Ca, and both cations are necessary element for growth of plants and animals [25]. They react with carbonate to produce water hardness [29]. The Ca⁺² and Mg⁺² minimum and maximum concentration of wastewater was 1.44- 4.60 and 0.1-0.42 mg.l⁻¹ respectively. Site 2 characterized by highest Ca⁺² value than other studied sites, while site 3 has more Mg⁺² concentration than others sites, with no significant differences (P≤0.05) between studied sites. These results are lower than noted by [5; 10; 20].

The site with the highest chloride levels is site 1 with mean value 59.7 mg.l⁻¹ followed by site 2 and 3 with 56.3 and 47.2 mg.l⁻¹ respectively. [30; 31] commented that human excreta increase the amount of Cl⁻ in sewage. Decrease in chloride concentration was observed from upstream to downstream. There is a significant difference (P \leq 0.05) between site 1 and site 3. These results come in accordance with that observed by [16]. According to [14] guidelines the water is entirely safe for irrigation.

The results of sulphate (SO_4^{-2}) showed increases in their values from site 1 (922 mg.l⁻¹) to site 2 (1001.6 mg.l⁻¹) to site 3 (1309 mg.l⁻¹). The quality of stream water is affected by sulphates. The concentration of sulphate exceed [22] maximum permissible level for irrigation purposes. Moreover, SO_4^{-2} concentration was always higher than those of Ca^{+2} .

Alkalinity is a measure of the buffering capacity of water [32]. Most alkalinity value is due to bicarbonate ion. Generally, low concentration of HCO_3 was recorded in Erbil wastewater channel with mean values of 0.089, 0.096 and 0.087 Meq.l⁻¹ respectively for studied sites. These values are lower than recorded by

previous studies on the same channel [5; 8; 16]. The water quality is non-restricted to use for irrigation according to guideline of [14]. High sulphate values in water caused Ca^{+2} precipitation which lead to relatively increase in Na⁺ contain in soil that adsorbs to clay surfaces and effects on nutrients balanced in plants [33].

Table 1: Physical-chemical properties of Erbil wastewater channel, data represented as mean± S.E. with minimum and maximum values, during studied period.

Variables	Sites							
	1	2	3					
pН	7.26 ± 0.065	7.51 ± 0.056	7.65 ± 0.072					
1	6.99- 7.78	7.15-7.77	7.24-7.98					
EC(µS.cm ⁻¹)	602.0±31.97	636.8±29.70	680.4±47.45					
(at 25°C)	486.15-838.8	517.9- 869.6	529.1- 1032.6					
TDS	345.13±21.88	380.14±21.25	661.46±272.07					
$(mg.l^{-1})$	236.76-491.2	264.18-529.9	265.56- 648.9					
Total hardness	277.91±11.22	287.08±13.17	296.33±8.78					
(mg CaCO ₃ .l ⁻¹)	226.0-346.0	240.0-380.0	242.0- 336.0					
Na ⁺	1.73 ± 0.16	2.07 ± 0.098	1.91 ± 0.077					
(Meq.l ⁻¹)	0.38-2.30	1.56-2.60	1.42-2.30					
(Meq.l ⁻¹) K ⁺	0.0053 ± 0.00044	0.0057 ± 0.00024	0.0051 ± 0.00048					
(Meq.l ⁻¹⁾	0.00- 0.01	0.00- 0.01	0.00- 0.01					
HCO ₃ ⁻	0.089 ± 0.0054	0.096 ± 0.0064	0.087± 0.0043					
(Meq.l ⁻¹) Ca ⁺²	0.70- 0.13	0.60- 0.14	0.70- 0.11					
Ca ⁺²	3.375±0.199	3.608 ± 0.170	3.525 ± 0.233					
(Meq.l ⁻¹)	1.80- 4.48	2.68-4.60	1.44- 4.32					
(Meq.l ⁻¹) Mg ⁺²	0.182 ± 0.024	0.178 ± 0.0154	0.200± 0.0126					
$\frac{(\text{Meq.I}^{-1})}{\text{SO}_4^{-2}}$	0.10- 0.42	0.10-0.29	0.12- 0.28					
SO4 ⁻²	19.21±2.233	20.87±1.875	27.27±3.598					
(Meq.l ⁻¹)	4.17-28.65	6.67-28.65	6.67-45.83					
СГ	1.655 ± 0.129	1.609 ± 0.073	1.350 ± 0.079					
(Meq.l ⁻¹)	0.86-2.68	1.26-2.23	0.88-1.77					
Na%	32.04 ± 2.405	35.48±1.583	34.39±1.373					
	11.63-41.43	28.29-47.22	27.59-45.15					
SAR	1.298 ± 0.120	1.521 ± 0.085	1.415 ± 0.050					
	0.32-1.84	1.14-2.16	1.14- 1.80					
Cd	25.70±2.475	17.13±1.529	16.35±1.071					
(µg.l ⁻¹)	0.00-46.73	0.00-28.04	0.00- 28.04					
Pb	15.44± 2.06	30.88± 4.31	50.74± 6.26					
(µg.l ⁻¹)	0.00- 61.76	0.00- 61.76	0.00- 179.41					
Cu	9.34± 1.81	0.00 ± 0.00	14.02 ± 1.44					
(µg.l ⁻¹)	0.00-18.69	0.00- 0.00	0.00- 18.69					
Zn	12.82±1.64	32.05 ± 4.84	25.64± 2.93					
(µg.l ⁻¹)	0.00-76.92	0.00-153.85	0.00- 76.92					

Sodium considered being the most important factor in determining the suitability of water for irrigation [34]. Maximum sodium percent (Na%) was 47.2% recorded in site 2. According to [35] the wastewater classified as good type of water for irrigation. These results come with results that obtained by [16], while, its lower than noted by [8] at the same polluted channel.

Generally, the mean concentrations of ions in wastewater were observed in the order: SO_4^{-2} > Cl> Na⁺> HCO_3> Ca⁺²> Mg⁺²> K⁺.

Water quality for Erbil wastewater channel characterized by dominance of the following heavy metals according to their content in sequences: Pb> Zn> Cd> Cu with highest mean values of 50.74, 32.05, 25.7 and 14.02 μ g.l-1 respectively. [10] found heavy metal dominance in the same polluted channel in the following order: Zn> Pb> Cu> Cd. From previous results its can be concluded that the water channel is polluted by heavy metals that comes from human activities and industrial effluent [36].

3.2 Data structure determination and source identification

PCA is a powerful pattern recognition technique that attempts to explain the variance of a large dataset of intercorrelated variables with a smaller set independent variables (Principal component) [37].

Principal component analysis/ factor analysis was performed on the normalized dataset (17 variables) for three different sites in twelve months, in order to identified important water quality parameters. An eigenvalue gives a measure of the significance of the factor: the factors with the highest eigenvalues are the most significant. Eigenvalues of 1.0 or greater are considered significant [38] classification of factor loading in thus "strong", "moderate" and "weak", corresponding to absolute loading values of > 0.75, 0.75- 0.5 and 0.5- 0.3, respectively [39]. Corresponding, variables loading and explained variance with rotated factors are presented in (Table 2).

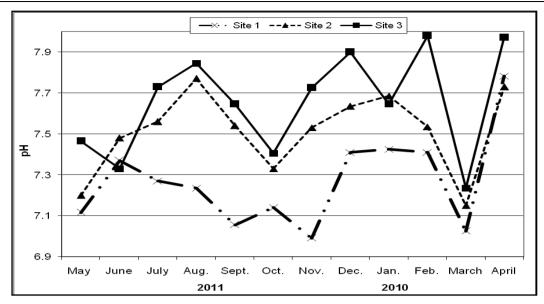


Figure 2: pH variation in Erbil wastewater channel during studied period.

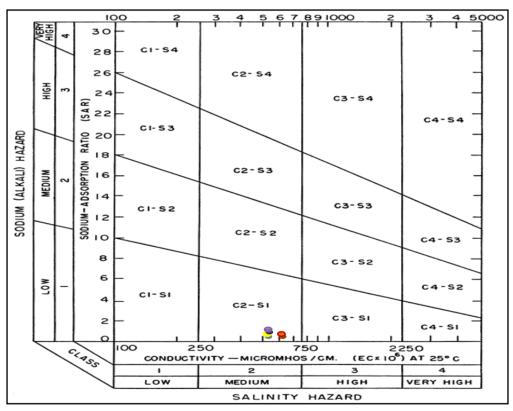


Figure 3: Classification of irrigation waters (19). Circles denoted the chemical data of the water sampling sites depending on mean value.

The results showed that first six components accounted for more than 80% of the total qualitative differences among the sites. The first component PC₁ accounted for 25.3% of the whole dataset, has strong positive loading on Na, Na% and SAR, which shows that the quality of water in these sites is affected by salinity factor, that may be comes from domestic uses and soap component [26]. For the second component PC₂ is associated with strong positive loading to total hardness and bicarbonate as the most significant parameters and moderate positive loading to Mg and Cl ions (Fig. 5 and 6). These variables are accounts for temporary hardness of water. While, third component PC₃ is strongly positive loading with SO₄⁻² and moderately positive loading with pH, EC, Pb, and negatively with K⁺ ions, it can be related to mineral salts and weathering [40].

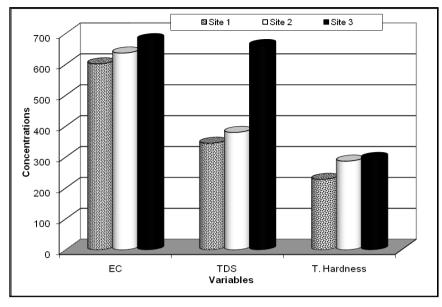


Figure 4 Shows increase in EC µS.cm⁻¹, TDS mg.l⁻¹ and Harness CaCO₃.l⁻¹ concentrations from upstream to downstream of Erbil wastewater channel.

3.3 Sites and variables similarity

The similarity between sites were obtained through cluster analysis using Wards method (linkage between groups) with Euclidian distance as a similarity measure and were amalgamated into dendogram plots (Fig. 7). As it shown there are similarity associations between site 1 and 2, both of them linked with site 3. Generally, in most studied variables site 2 considered as intermediate stage between sites 1 and 3 in both increasing value of variables or decreasing from upstream of channel toward the downstream.

Fig. (8) showed the similarity degree between studied variables in which two clusters of three groups were formed. First group linked TDS, EC and total hardness. The higher concentration of TDS related to EC and hardness ions is an indication of man- made pollution due to domestic waste [41]. Second group association between SO_4^{-2} , Na%, Cd and Pb were obtained. While, third group consists of association between three subgroups; in which association found between K⁺, HCO₃, Mg⁺² with Cl, SAR, Na⁺, pH, Ca⁺², Cu all with Zn. These ions suggest the associated with a combination of various hydrogeochemical processes and augment more mineralized water [24].

IV. Conclusion

Hydrogeochemical, mineralization and sewage effluent discharges are the most important sources of ions in the polluted channel in which increases in ions concentrations from upstream polluted channel towards downwards according to EC, TDS and total hardness values were observed during period of investigation. Most of ions concentration comes under safeguard public health for irrigation purposes, but other human public health factors must be taking account such as pathogenic microrganisms.

References

- [1]. E.O. Igbinosa, and A.I. Okoh, Impact of discharge wastewater effluent on the physic- chemical qualities of a receiving watershed in a typical rural community, *International Journal of Environment Science and Technology*, *6*(2), 2009, 175-182.
- [2]. K. Šekabira, H. O. Origa, T.A. Basamba, G. Mutumba, and A. Kakudidi, Heavy metal assessment and water quality values in urban stream and rain water, *International Journal of Environment Science and Technology*, 7(4), 2010, 759-770.
- [3]. B.Y. Mustafa, and S. Sabir, Reuse of Erbil City sewage for irrigation purposes, *Proc. Scientific conf. of water- Erbil. Brayeti-Center, 18,* 2001, 303-317.
- [4]. A.H. Pejman, G.R. Nabi Bidhendi, A.R. Karbassi, N. Mehrdadi, and M. Esmaeili Bidhendi, Evaluation of spatial and seasonal variations in surface water quality using multivariate statistical techniques. *International Journal of Environment Science and Technology*, 6(3), 2009, 467-476.
- [5]. Y.A. Shekha, An ecological study on the main sewage channel of Erbil city, M.Sc. Thesis, University of Salahaddin -Erbil, Iraq, 1994.
- [6]. F.H. Aziz, S.N. Darokha, and Y.A. Shekha, Ecological and microbiological studies of Erbil city's sewerage. Proc. Scientific conf. of water- Erbil. Brayeti-Center, 18, 2001, 189-206.
- [7]. L.A. Ali, Algal studies in sewage water within Erbil city, M.Sc. Thesis. University of Salahaddin Erbil, Iraq, 2002.

Table 2:	Eigenvalue and percentage of variance explained by each of the seven principal components
	(PCs) for Erbil wastewater channel studied variables.

Principal component	PC ₁	PC ₂	PC ₃	PC ₄	PC ₅	PC ₆		
Eigenvalue	4.5590	2.9380	2.5070	1.8890	1.329	1.224		
% total variance explained	25.326	16.320	13.930	10.496	7.381	6.798		
% cumulative variance	25.326	41.646	55.576	66.072	73.453	80.251		
Rotated factor correlation coefficients								
рН	-0.269	-0.071	0.591	0.056	-0.401	-0.172		
EC	0.354	-0.247	0.600	0.496	-0.009	0.162		
TDS	0.028	-0.124	-0.155	-0.465	0.418	0.045		
Total hardness	0.010	0.899	0.302	0.113	-0.030	0.104		
Na ⁺	0.897	0.208	0.025	0.359	0.025	0.006		
K ⁺	0.447	0.452	-0.608	0.131	-0,218	-0.055		
HCO ₃	-0.040	0.825	-0.286	0.100	-0.034	-0.065		
Ca ⁺²	-0.094	0.333	0.050	0.871	0.191	0.055		
Mg^{+2}	0.078	0.609	0.214	-0.708	-0.162	0.063		
SO ₄ ⁻²	0.063	0.087	0.790	-0.012	-0.101	-0.081		
СГ	0.328	0.673	-0.336	0.008	0.278	-0.066		
Na%	0.966	0.010	-0.049	-0.171	-0.084	-0.006		
SAR	0.990	0.061	-0.001	0.038	-0.056	-0.024		
Cd	-0.213	0.021	0.104	0.193	0.770	-0.093		
Pb	0.002	-0.032	0.716	0.050	0.306	0.091		
Cu	0.117	-0.135	-0.075	0.043	-0.369	0.785		
Zn	0.230	-0.169	-0.055	0.022	-0.393	-0.710		

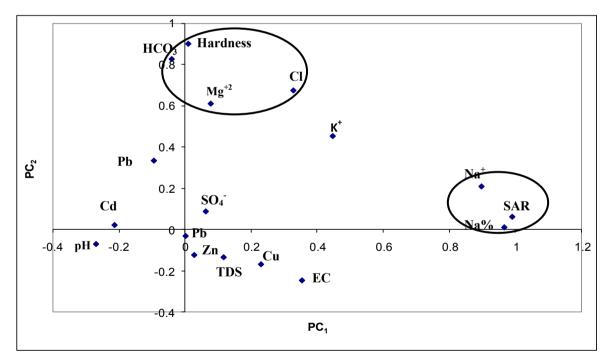


Figure 5: Principal component analysis (PCA) scatterplot for Erbil wastewater samples on the basis of studied water variable characteristics.

- [8]. D.G.A. Ganjo, F.H. Aziz, and Y.A. Shekha, An attempt for reuse of the wastewater of Erbil city for irrigation purposes. *Zanko Journal*, 18(2), 2006, 148-175.
- [9]. B.A.A. Sofi, An ecological study on the main sewage channel of Erbil city, with particular reference to self-purification, M.Sc. Thesis, University of Salahaddin Erbil, Iraq, 2007.
- [10]. M.H.H. Lak, Environmental study of Arab-kand wastewater channel in Erbil governorate, Kurdistan region, Iraq, M.Sc. Thesis, University of Salahaddin -Erbil, Iraq, (2007).

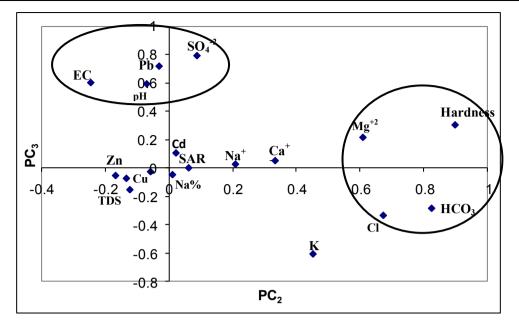


Figure 6: Principal component analysis (PCA) scatterplot for Erbil wastewater samples on the basis of studied water variable characteristics.

- [11]. Y.A. Shekha, *The effect of Erbil city wastewater discharge on water quality of Greater Zab River, and the risks of irrigation*, doctoral diss., Baghdad University, Iraq, 2008.
- [12]. Y.A. Shekha, N.H. Hayder, and Y.O. M. Al-Barzingy, The effect of wastewater disposal on the water quality and phytoplankton in Erbil wastewater channel, *Baghdad Science Journal*, *7(2)*, 2010, 984-993.
- [13]. American Public Health Association (A.P.H.A.), Standard methods for the examination of water and wastewater (1015 Fifteenth Street, NW, Washington, DC., 1998).
- [14]. M.B. Pescod, Wastewater treatment and use in agriculture (FAO irrigation and drainage paper 47, Rome, Italy, 1992).
- [15]. R. S. Ayers, and D. W. Westcot, *Water Quality for Agriculture* (FAO Irrigation and Drainage Paper 29, revision 1, Rome, Italy, 1985).
- [16]. K.N. Amin, and S.Q. Aziz, Feasibility of Erbil wastewater reuse for irrigation, Zanco Journal, 17(2), 2005, 63-77.
- [17]. H.L. Golterman, R.S. Clymo, and M.A.M. Ohnstand, Methods for physical and chemical analysis of freshwater (Blackwell Science Publication, NO.8, Oxford, 1978).
- [18]. H.H. Rump, Laboratory manual for the examination of water, wastewater and soil (Wiely-VCH Verlag Publication, Germany, 1999).
- [19]. L.A. Richards, *Diagnosis and improvement of saline and alkali soils* (Agriculture hand book, US Department of Washington, 1969).
- [20]. A. Bahri, Fertilizing value and polluting load of reclaimed water in Tunisia, Water Research, 32(11), 1998, 3484-3489.
- [21]. M.J. Hammer, Water and wastewater technology (Prentice Hall International, Inc. USA, 1996).
- [22]. D.K. Todd, Ground water hydrology (John Wiley and Sons, Inc. USA, 1980).
- [23]. M.M. Duqqah, Treated sewagewater use in irrigated agriculture: Theorical design of farming systems in Seil Al Zarqa and the middle Jordan valley in Jordan, doctoral diss., Wageningen University, Wageningen, 2002.
- [24]. V. Bhardwaj, D.S. Singh, and A.K. Singh, Water quality of the Chhoti Gandak river using principal component analysis, Ganga plain, India, *Journal Earth System Science*, 119(1), 2010, 117-127.
- [25]. J.D. Hem, Study and interpretation of the chemical characteristics of natural water (USGS water supply, 1985).
- [26]. R.G. Wetzel, *Limnology* (W.B. Saunders Company, 1975).
- [27]. S.N. Davis, and R.J. Dewiest, Hydrogeology (John Wiley and Sons, Inc.USA, 1966).
- [28]. Z. Goklap, I. Tas, F. Ozkay, and S. Akgul, Wastewater use in irrigation- Ankara case study, African Journal of Agricultural Research. 6(7), 2011 1807-1812.
- [29]. J. Rosozka, Euphrates and Tigris, Mesopotamin ecology and destiny, (Monographic Biologica, The Hageue Boston, UK, 1980).
- [30]. C.N. Sawyer, and P.L. McCarthy, *Chemistry for environmental engineering* (McGraw-Hill Book Company, Singapore, 1978).
- [31]. F.W. Ntengwe, Pollutants loads and water quality in streams of heavily populated and industrialized towns. *Physics and Chemistry* of the Earth. 31, 2006, 832-839.
- [32]. G. Kiely, *Environmental engineering* (McGraw-Hill Company, 1997).
- [33]. U.A. Kovda, B. Yaron, and J. Shalhavet, Quality of irrigation water (FAO, Roma, 1973).
- [34]. J.D. Rhoades, and S.D. Merill, Assessing the suitability of water for irrigation: Theoritical and empirical approaches (Agriculture Research, US Department of Agriculture, USSI, 1975).
- [35]. L.V. Wilcox, Classification and use of irrigation waters (US Department of Agriculture, 1955).
- [36]. U.K. Pradhan, and P.V. Shirodkar, Assessment of the impact of developmental activities on estuarine environments of Mandovi and Zuari rivers of Goa along the west coast of India, *Proceedings of International Conf. in Ocean Engineering*, 1-5 Feb. 2009. IIT Madaras, Chennai, India, 1145-1156).
- [37]. J.A. Simeonov, J.A. Stratis, C. Samara, G. Zachariaadis, D. Voutsa, A. Anthemidis, M. Sofoniou, and Th. Kouimtzis, Assessment of the surface water quality in northern Greece, *Water Research*, 37, 2003, 4119-4124.
- [38]. S. Shrestha, F. Kazama, Assessment of surface water quality using multivariate statistical techniques: A case study of the Fuji river basin, Japan, *Environmental Modeling. Software*, 22(4), 2007, 464-475.

- [39]. C.W. Liu, K.H. Lin, and Y.M. Kuo, Application of factor analysis in the assessment of groundwater quality in a Blackfoot disease area in Taiwan, *Science Total Environment*, 313(1-3), 2003, 77-89.
- [40]. S. Shrestha, F. Kazama, and T. Nakamura, Use of principal component analysis, factor analysis and discriminate analysis to evaluate spatial and temporal variations in water quality of the Mekong river, *Journal of Hydroinformatics*, 10(1), 2008, 43-56.
- [41]. C.W. Fisher, and W.F. Mullican, Hydrochemical evolution of sodium- sulfate and sodium- chloride ground water beneath the northern Chihuahuan deseart, Trans- Pecos, Texas, USA. *Hydrogeology Journal*, *5*, 1997, 4-16.

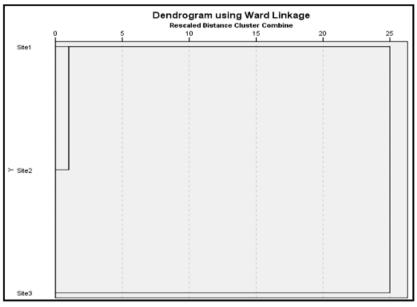


Figure 7: Similarity <u>dendogram</u> among sampling sites along Erbil wastewater channel from cluster analysis.

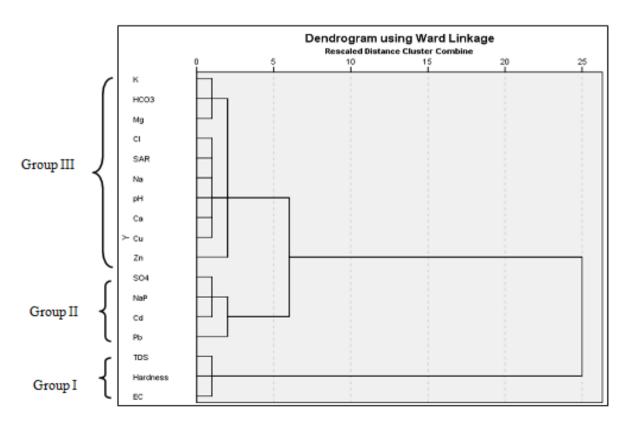


Figure 8 Similarity dendogram among variables concentrations of Erbil wastewater effluent from cluster analysis.