L-Ascorbic acid and α-Tocopherol to protect against arsenic trioxide induced oxidative stress in H9c2 cardiomyocytes

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Abstract: Arsenic has been used as a therapeutic agent against various diseases right from the historical periods. Arsenic trioxide (As₂O₃), a first line chemotherapeutic against Acute Promyelocytic Leukemia (APL), is well known for its cardiotoxicity. We hypothesized that combination of drug and protective molecules can be used for reducing the side effects of chemotherapeutics. The present study evaluates the protective role of L-ascorbic acid (L-AA) and α-Tocopherol (α-TOC) against oxidative stress in H9c2 cardiomyocytes induced by As₂O₃. Clinically relevant concentration of arsenic trioxide induces cytotoxicity in cardiac cells. Treatment with antioxidant vitamins significantly reduced the cell death and apoptosis in H9c2 cells. The activity of antioxidant enzymes such as catalase and the concentration of reduced glutathione, which was lowered due to As₂O₃ treatment, were found to be increased in cells co treated with L-AA and α-TOC. The treatment with antioxidant vitamins also reduced lactate dehydrogenase release in cells. The overall results clearly indicate the protective potential of L-AA and α-TOC against arsenic induced chemotherapeutic stress on cardiomyocytes.

Keywords: Arsenic trioxide, Cardiotoxicity, H9c2 cardiomyocytes, L-Ascorbic acid, Oxidative stress, α-Tocopherol

I. Introduction

Acute promyelocytic leukemia (APL) is a distinctive subtype of acute myelogenous leukemia (AML) and accounts for approximately 10% to 15% of adult myeloid leukemia.[1] APL is a disease in which cancer cells are found in blood and bone marrow. Treatment for leukemia is complex and depends on a number of factors including the type of leukemia, the extent of the disease and whether the leukemia has been treated before. Arsenic is an environmental toxin but it has also been used in medicine as a therapeutic agent for more than 2400 years.[2] Inorganic arsenic is now a first choice cancer chemotherapeutic against certain leukemia.[3],[4] The US Food and Drug Administration (FDA) approved the use of Trisenox brand arsenic trioxide (As₂O₃) for the treatment of acute promyelocytic leukemia in September 2000.[5]

Most APL cases are characterized by a t(15;17) translocation that fuses the promyelocytic leukemia (PML) gene on chromosome 15 to the retinoic acid receptor α (RARα) gene on chromosome 17, resulting in the formation of the oncoprotein promyelocytic leukemia/retinoic acid receptor (PML-RARα) fusion protein. Arsenic induces differentiation through degradation of the fusion protein.[6],[7],[8] Due to its toxic nature, arsenic trioxide carries significant health risks. It can cause QT interval prolongation which can lead to torsades de points and sudden death. Eliminating the cardiotoxicity induced by arsenic trioxide is expected to have a tremendous impact on the treatment of acute promyelocytic leukaemia. Studies showed that arsenic trioxide (As₂O₃) induces cell death in cancer cells by the generation of reactive oxygen species, cell cycle arrest, and activation of apoptotic pathways.[9],[10] The same mode of action in normal cells causes cell toxicity which is fatal for cellular survival.

Various antioxidant molecules are found to be effective in countering free radical induced damage that occur due to arsenic chemotherapy.[11] L-Ascorbic acid (L-AA) or Vitamin C, a low molecular weight, water-soluble antioxidant vitamin, possesses pro oxidant properties that helps the body’s own free radical defense mechanism and destroy tumors in their initial stages. Evidences showed that ascorbic acid can improve mitochondrial function by improving the thiol status; thereby preventing reactive oxygen species – mediated damage. Ascorbic acid is not likely to increase the antineoplastic activity of arsenic but could provide protection to both normal and neoplastic cells.[2]

α-Tocopherol or Vitamin E, a lipid soluble vitamin, react with free radicals and ROS. The administration of α-Tocopherol will decrease extend of lipid peroxidation; a major contributor to atherosclerosis and cardiovascular diseases. It can also function as a peroxyl radical scavenger that terminates chain reaction.[12],[13] α-Tocopherol has been shown to protect against arsenic intoxication.[14]

In the present study we are evaluating the ameliorative effect of antioxidant vitamins against the oxidative stress induced by As₂O₃ in H9c2 cardiomyocytes.
II. Materials And Methods

2.1 Reagents

Arsenic trioxide (As$_2$O$_3$), L-Ascorbic acid (L-AA), α-Tocopherol (α-TOC), Acridine orange and Ethidium Bromide were purchased from Sigma (USA). Fetal Bovine Serum (FBS) was purchased from Invitrogen. 3-(4, 5, dimethylthiazol-2-yl)-2, 5- diphenyl tetrazolium bromide (MTT), dimethyl sulfoxide (DMSO), Dulbecco’s modified Eagle’s medium (DMEM), Trypsin - EDTA solution and other chemicals were purchased from Himedia Pvt Ltd (Mumbai, India).

2.2 Cell culture and treatment

H9c2 cardiomyocytes were obtained from National Centre for Cell Science (NCCS), Pune, India. Cells were cultured in Dulbecco’s Modified Eagle’s Medium (DMEM) supplemented with 10% FBS and incubated at 37°C in humidified atmosphere in a 5% CO$_2$ incubator. Cells were passaged regularly and sub cultured to 80% confluence before the experiments. The experimental group consists of

(a) Control cells.
(b) Cells treated with 0.2% ethanol (Negative control) for 48 hours.
(c) Cells treated with 10 µM As$_2$O$_3$ for 48 hours.
(d) Cells treated with 100 µM L-Ascorbic acid for 48 hours.
(e) Cells treated with 50 µM α-Tocopherol for 48 hours.
(f) Cells treated with 10 µM As$_2$O$_3$, 100 µM L-Ascorbic acid and 50 µM α-Tocopherol for 48 hours.

2.3 Morphological Analysis

After 48 hours of treatment with As$_2$O$_3$, L-AA and α-TOC, the morphology of H9c2 cardiomyocytes were observed using an inverted phase contrast microscope (Olympus CKX41 with Optika Pro5 camera) at 10 x magnification.

2.4 Cell viability assay

Cell viability of various experimental groups was determined after 48 hours of incubation by MTT assay. MTT is a colorimetric assay that measures the reduction of yellow 3-(4,5 dimethythiazol-2-yl)-2, 5- diphenyl tetrazolium bromide (MTT) by mitochondrial succinate dehydrogenase. The MTT enters the cells and passes into the mitochondria where it is reduced to an insoluble, coloured formazan product. The cells are then solubilised with an organic solvent dimethyl sulfoxide and the released, solubilised formazan product was measured using an ELISA plate reader (Erba Manheim, Germany) at 540 nm.[15]

2.5 Lactate Dehydrogenase Release Assay

Lactate dehydrogenase (LDH) is used as a quantitative marker enzyme for the intact cell. Measurement of lactate dehydrogenase release is an important and frequently applied test for cell damage. LDH leakage assay was performed with cell free supernatant mixed with potassium phosphate buffer, 6mM NADH solution and sodium pyruvate solution. The OD was recorded at 340nm in a spectrophotometer.[16]

2.6 Lipid peroxidation Assay

Lipid peroxidation in cell cultures was estimated by measuring the formation of malondialdehyde (MDA). The treated cells were centrifuged at 4000 rpm for 10 minutes. Cell lysis buffer was added to the pellet and kept for 30 minutes incubation at 4°C. 70% alcohol and 1% TBA were added to each of the tubes and were kept in a boiling water bath for 20 minutes. Acetone was added to all the test tubes after cooling to room temperature and the absorbance was read at 535 nm.[17]

2.7 Fluorescent microscopy Observations

Cells were labeled with Acridine Orange/ Ethidium Bromide (AO/EB) to detect apoptosis. A mixture of AO (100 µg/ml) and EB (100 µg/ml) were added and incubated at room temperature for 10min. After washing with PBS the cells were observed under a fluorescent microscope (Olympus CKX41 with Optika Pro5 camera).[18]

2.8 Determination of antioxidant enzyme activities

Activities of antioxidant enzymes like reduced glutathione (GSH) and Catalase (CAT) were assayed according to the method of Moron et al.[19] and Sinha et al.[20] respectively.

2.9 Estimation of Nitrate by Griess method
H9c2 cells were plated in a 96-well plate (1×10^5 cells/well), and incubated with appropriate concentrations of As₂O₃, L-AA and α-TOC for 48 hours. The concentration of nitrate was determined using Griess reagent (1% sulphanilamide, 0.1% naphthylethylene diamine dichloride and 2% phosphoric acid). The amount of nitrate present in various samples was measured at 540nm.[21]

2.10 Statistical Analysis
Data were collected from repeated experiments and the results were presented as mean (± Standard deviation). Data were subjected to one-way analysis of variance (ANOVA) using Origin, version 7, Origin Lab Corporation, Northampton, USA. P<0.05 was considered to be statistically significant.

III. Results

3.1 Effect of Arsenic trioxide on cell morphology:
Fig 1B. showed that treatment with As₂O₃ (10 μM) induced changes in the morphology of H9c2 cells such as shrinkage, rounding up detachment from the plate when compared with the normal cells (Fig 1A). Combination treatment using L-Ascorbic (100 μM) and α-Tocopherol (50 μM) were found to be effective in maintaining the normal cell morphology (Fig 1F, 1G, 1H).

3.2 Effect of L- Ascorbic Acid and α-Tocopherol on arsenic trioxide induced cytotoxicity in H9c2 cells.
As shown in Fig 2., 10 μM As₂O₃ reduced the viability of H9c2 cells after 48 hours of treatment. An increase in the percentage of viability was observed in cells treated with a combination of As₂O₃ and L-AA and As₂O₃ and α-TOC. Combined treatment with 100 μM L-Ascorbic acid and 50 μM α-Tocopherol with As₂O₃ significantly increased the cell viability when compared with all other groups (Fig. 2).

3.3 L- Ascorbic Acid and α-Tocopherol reduces LDH release in Arsenic trioxide-treated H9c2 cells

![Graph showing cell viability and LDH release](Graph.png)
The cellular LDH release was markedly increased in cells treated with As$_2$O$_3$ (10 μM) whereas combined treatment with L-Ascorbic acid (100 μM) and α-Tocopherol (50 μM) with As$_2$O$_3$ significantly reduced the LDH leakage (Fig. 3).

![Fig 3. LDH leakage](image)

**Fig 3. LDH leakage.** Data represented as mean ±SD, *p<0.05 versus normal control and ^p<0.05 versus As$_2$O$_3$ treated groups.

### 3.4 L-Ascorbic acid and α-Tocopherol lowers lipid peroxidation in As$_2$O$_3$ treated cells

A high level of lipid peroxidation was observed in H9c2 cells treated with 10 μM As$_2$O$_3$. In the presence of 100 μM L-Ascorbic acid and 50 μM α-Tocopherol, the malondialdehyde (MDA) release was markedly reduced. This indicates that administration of antioxidant vitamins significantly reduced the extent of lipid peroxidation caused by arsenic trioxide. (Fig. 4).

![Fig 4. Lipid peroxidation assay](image)

**Fig 4. Lipid peroxidation assay.** Data represented as mean ±SD, *p<0.05 versus normal control and ^p<0.05 versus As$_2$O$_3$ treated groups.

### 3.5 As$_2$O$_3$ induced apoptosis can be reduced by co-treatment with L-Ascorbic acid and α-Tocopherol

The cells were observed under a fluorescent microscope at 10x magnification. Viable cells appeared as green and apoptotic cells as orange. Morphological changes such as cell shrinkage and nuclear condensation were observed in the cells treated with arsenic trioxide when compared with the control cells (Fig. 5A, 5C). AO/EB staining showed that combination treatment using antioxidant vitamins such as L-Ascorbic acid and α-Tocopherol was effective to protect the cells from apoptosis (Fig. 5F, 5G, 5H).
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Fig 5. Morphological observation of H9c2 cells by AO/EB staining. (A) Control cells (B) Cells treated with 0.2% ethanol (C) Cells treated with 10μM As2O3 (D) Cells treated with 100μM L-AA (E) Cells treated with 50μM α-TOC (F) Cells treated with 10μM As2O3 and 100μM L-AA (G) Cells treated with 10μM As2O3 and 50μM α-TOC (H) Cells treated with 10μM As2O3, 100μM L-AA and 50μM α-TOC (original magnification x 10).

3.6 Antioxidant Status

The activity of reduced glutathione (GSH) was found to be significantly reduced by treatment with 10 μM concentration of As2O3 when compared to the untreated cells. Co-treatment with 100 μM L-AA and 50 μM α-TOC significantly reduced the inhibition of GSH activity caused by arsenic trioxide (Fig. 6A). Similarly, the catalase (CAT) activity was also inhibited significantly by As2O3 (Fig. 6B). The combination treatment with L-AA and α-TOC in these cells also reduced the catalase inhibition by arsenic trioxide treated cells.

Fig 6A. Effect of arsenic trioxide, L-Ascorbic acid and α-Tocopherol on reduced glutathione level. Data represented as mean ±SD, *p<0.05 versus normal control and #p<0.05 versus As2O3 treated groups.

Fig 6B. Effect of arsenic trioxide, L-Ascorbic acid and α-Tocopherol on catalase activity. Data represented as mean ±SD, *p<0.05 versus normal control and #p<0.05 versus As2O3 treated groups.

3.7 Determination of Nitrate levels

The nitrate levels of various experimental groups were estimated using Griess reagent. The amount of nitrate released was found to be significantly higher in Arsenic trioxide treated cells when compared with the control cells. Fig 7 showed that the combination of L-AA and α-TOC with arsenic trioxide significantly reduced the level of nitrate released.
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The current study has shown the protective efficacy of L-Ascorbic acid and α-Tocopherol against arsenic trioxide induced oxidative stress in H9c2 cardiomyocytes. H9c2 is a clonal cell line derived from fetal rat heart and is morphologically similar to embryonic cardiomyocytes.[22]

Exposure of clinically relevant concentration of As₂O₃ induces apoptosis in H9c2 cardiomyocytes. The mechanism by which As₂O₃ induces cell death is the favorable chemical reaction between arsenic and thiol groups within a protein. The valence orbitals of arsenic have a better affinity to sulfur than with oxygen, leading to As-thiolate bond formation with the release of water.[23] It is believed that low or physiologic doses of L-Ascorbic acid were not cytotoxic, and is beneficial in both cardiovascular diseases and cancer.[24] The most active form of vitamin E, α-Tocopherol is an antioxidant which prevents the peroxidation of unsaturated fatty acids by scavenging lipophilic radicals within membranes.[25] These compounds have cardioprotective and hepatoprotective roles also. Thus, our study focuses on the protective efficacy of these antioxidant vitamins against As₂O₃-induced toxicity and oxidative stress in H9c2 cells.

Results from the MTT, LDH and lipid peroxidation assays showed that As₂O₃ is cytotoxic to H9c2 cardiomyocytes. In addition to these, morphological alterations due to As₂O₃ such as cell surface shrinkage followed by rounding up occur as a result of membrane damage. Reduction in the viability of cells was also observed. A combination of L-AA (100μM) and α-TOC (50μM) was found to be effective against cytotoxicity at 10 μM concentration of As₂O₃. Arsenic trioxide induces myocardial membrane damage which leads to an increase in the LDH leakage as indicated in the current study. A marked increase in the level of lipid peroxidation was observed in arsenic trioxide treated cells. As₂O₃ causes the formation of free radicals which directly attack the poly unsaturated fatty acids (PUFA) present in the cell membrane resulting in the formation of MDA which was biochemically quantified by lipid peroxidation assay. We observed a significant reduction in the LDH activity and lipid peroxidation in cells co-treated with L-Ascorbic acid and α-tocopherol which may be due to the membrane protective activity exerted by these antioxidant vitamins.

Studies showed that apoptosis plays an important role in the progression of heart failure.[26] The present study indicates As₂O₃ induced apoptosis in cardiomyocytes, as observed from the AO/EB fluorescent double staining. The evaluation of antioxidant enzymes like GSH and CAT in arsenic trioxide-treated cells were found to be significantly lower. High levels of GSH are associated with cellular resistance to arsenic and decreasing intracellular GSH concentrations cause increased sensitivity to As₂O₃. GSH functions as the major auto-oxidant of the cells, scavenging free radicals and to inactivate toxins and chemotherapeutics.[27] Increased MDA production directly indicates a reduction in GSH Levels which results in elevated ROS production resulting in lipid peroxidation leading to membrane damage of cells. L-AA and α-TOC was able to protect the enzymes from depletion by As₂O₃, mainly due to the antioxidant property exhibited by these vitamins.

The formation of reactive oxygen species and nitrogen species by As₂O₃ is one of the most studied mechanisms for arsenic induced toxicity.[28] An increased level of nitrate, a stable product of Nitrile Oxide (NO) was also detected in As₂O₃ treated cells. This may play a major role in As₂O₃ induced apoptosis in H9c2 cardiomyocytes. NO is a highly unstable free radical produced by the enzyme Nitric Oxide Synthases (NOS).[29] NO binds to cytochrome oxidase and can block the mitochondrial respiratory chain thereby induces apoptosis. It is believed that antioxidant vitamins such as L-AA and α-TOC is beneficial in both cardiovascular diseases and cancer.[24] We suggest that these antioxidant vitamins reduces the cytotoxicity of As₂O₃ by quenching intracellular ROS and nitrate levels thereby reducing the level of lipid peroxidation and apoptosis.[30]
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V. Conclusion

The present study evaluates the protective effect of L-ascorbic acid and α-Tocopherol against arsenic trioxide induced toxicity in H9c2 cardiomyocytes. From the current study, it is evident that co-treatment with antioxidant vitamins protected the cells from the toxic effects of arsenic trioxide. In conclusion, using an in vitro model of H9c2 cells, we suggested that these antioxidant vitamins may be a valuable option for reducing the cardiac toxicity induced by arsenic trioxide thereby rising its effectiveness as a chemotherapeutic agent.

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References

[19]. Moron MS, Depierre JW, Mannervik B. Levels of glutathione, glutathione reductase and glutathione S-transferase activities in rat lung and liver. Biochim Biophys Acta. 582(1): 67-78.