

Simultaneous determination of paracetamol and diphenhydramine hydrochloride mixture in the presence of their degradation products

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Abstract: New accurate, selective, sensitive and precise methods were developed and validated for determination of paracetamol and diphenhydramine hydrochloride in the presence of P-amino phenol, the hydrolytic degradate and the most potential impurity of paracetamol and the N oxide degradation product of diphenhydramine in bulk form and in pharmaceutical formulation. Method A uses double divisor second derivative of ratio spectrophotometric technique, at 304nm for paracetamol and 256.4nm for diphenhydramine hydrochloride. Method B utilizes Principle Component Regression (PCR) and Partial Least Squares (PLS) chemometric techniques for quantification of the four components using a UV spectrum range of 210-350 nm. The proposed methods were successfully applied to the analysis of the mentioned drugs either in bulk powder or in pharmaceutical formulation without interference from other dosage form additives, and the results were statistically compared with the pharmacopoeial method.

Keywords: Paracetamol, diphenhydramine hydrochloride, P-amino phenol, double divisor spectrophotometric method, multivariate spectral analysis.

I. Introduction

Paracetamol (PC) is N-(4-hydroxy phenyl) acetamide. It is Para- aminophenol derivative; it has analgesic and anti-pyretic properties [1]. Diphenhydramine hydrochloride (DH) is 2-Diphenyl methoxy-N,N-dimethylethanamine, it has antihistaminic action and used for symptomatic relief of hypersensitivity reactions [2]. Their structures are shown in "Fig.1".

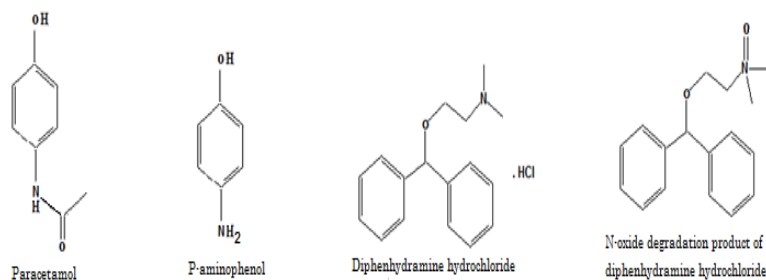


figure 1. structures of the studied drugs.

PC and DH are co-formulated in Panadol night tablets for temporary relief of pain associated with sleeping disorders. Simultaneous determination of the two drugs is time saving and less tedious than the determination of the two drugs by separate methods. The U.S. Pharmacopeia (USP 2007) has described a monograph to determine the binary mixture of PC and DH, depending on RP-HPLC using C₁₈ column and phosphate buffer- acetonitrile in the ratio of 94:6 v/v as a mobile phase with UV detection at 225 nm [3].

Few analytical methods have been described for simultaneous determination of PC/DH binary mixture including HPLC methods [4-6], NIR spectroscopy [7] and multivariate spectrophotometry [8]. A single published literature described the simultaneous quantification of PC and DH in presence of P-amino phenol (PAP); by first derivative (¹D) spectrophotometric, the first derivative of ratio spectra (¹DD), multivariate spectrophotometric and HPTLC-densitometric methods, in bulk material and pharmaceutical preparations [9].

Therefore the aim of this work is to describe a simple, sensitive, precise, accurate and time saving method for the simultaneous determination of PC and DH in the presence of both PAP and the N oxide

degradation product of DH, Diphenylmethoxy-N,N-dimethyl ethanamine N-oxide (NOD), which can be used in pure forms or in pharmaceutical formulation.

II. Experimental

2.1. Materials

2.1.1. Pure standards.

Paracetamol and Diphenhydramine Hydrochloride were kindly supplied by GlaxoSmithKline Company Egypt. Their purity was found to be $100.11 \pm 1.197\%$ and $99.96 \pm 0.743\%$, respectively according to the official HPLC method [3].

2.1.2. Pharmaceutical formulation.

Panadol night® tablets (Batch No. 124992) labeled to contain 500 mg of Paracetamol and 25 mg of Diphenhydramine Hydrochloride, manufactured by GlaxoSmithKline company.

2.1.3. Chemicals and reagents.

All reagents and chemicals used throughout this work were of analytical grade and were used without further purification. Methanol was of HPLC grade and obtained from ChromasolvW, Sigma-Aldrich Chemie GmbH, Germany. Ethyl acetate was from E. Merck, Germany, Hydrogen peroxide solution, sodium lauryl sulphate and acetone and acetic acid were from Al-Nasr pharmaceutical chemicals company, Abu Zaabal, Cairo, Egypt. P-aminophenol was from Riedel-dehaen-AG- Germany; its purity was certified to be 99%.

2.1.4. Preparation of N oxide degradation product (NOD).

A 0.2g aliquot of DH powder was mixed with 5mL of 1 M sodium hydroxide solution to liberate the free base which was then extracted by ether followed by spontaneous vaporization of the solvent. 5 mL of 30% hydrogen peroxide solution is added to the free base and covered carefully. Six hours were sufficient for complete oxidation of the free base. The reaction progress was followed via TLC using ethyl acetate - acetone - methanol - acetic acid - sodium lauryl sulphate (5:5:1:0.25:0.2, v/v/v/v/w) as a developing system. After complete oxidation, the solution was left to dry at room temperature and the degradation product powder was identified by IR and mass spectrometry, and used for preparation of the stock solution of the degradation product.

2.2. Instruments

A UV lamp with short wavelength 254nm (USA) and a double beam UV-visible spectrophotometer (SHIMADZU, Japan) model UV-1601 PC with quartz cells of 1 cm path length, connected to IBM compatible computer. The software was UVPC personal spectroscopy software version 3.7. Data analysis was performed using PLS-Toolbox 2.0 running under MATLAB, version 6.5.

2.3. Prepared solutions

2.3.1. Standards solutions For stock standard solutions of PC, DH, PAP and NOD (1 mg mL^{-1} in methanol), 0.1 g of each of the four components were accurately weighed into four separate 100 mL volumetric flasks; 50 mL of methanol was added to each flask and shaken to dissolve, then the volume was made up to the mark with methanol. For working standard solutions of PC, DH, PAP and NOD ($100 \text{ } \mu\text{g mL}^{-1}$), 10 mL of each stock standard solutions (1 mg mL^{-1}) were transferred accurately into four separate 100 mL volumetric flasks, and then the volume was completed to the mark with methanol.

2.3.2. Laboratory prepared mixtures

Mixtures containing different ratios of PC, DH, PAP and NOD were prepared using their respective working solutions in methanol.

2.4. Methodology

2.4.1. Double divisor second derivative of ratio ²DD spectrophotometric method

2.4.1.1. Linearity and construction of calibration curves

The ²DD spectra of PC and DH in a concentration range of 2-45 $\mu\text{g mL}^{-1}$ for each drug were recorded using a mixture of PAP and NOD containing $10 \mu\text{g mL}^{-1}$ of each as a divisor, $\Delta\lambda=8$ and scaling factor =500. The peak amplitudes were measured at 304nm for PC and at 256.4nm for DH.

The calibration curves were constructed relating the peak amplitudes to the corresponding concentrations for each drug. The regression equations were computed.

2.4.1.2. Analysis of laboratory prepared mixtures of PC, DH, PAP and NOD

Mixtures containing different concentrations of PC, DH, PAP and NOD were analyzed by applying the above-described procedure "Table 1".

2.4.2. Chemometric methods

2.4.2.1. Construction of the training set

Five level four factor calibration design was used [10]. Different mixtures containing different ratios of PC, DH, PAP and NOD were prepared “Table 2, mixtures 1-13”.

The absorption spectra of the mixtures were recorded and transferred to MATLAB for data manipulation. Thirteen mixtures were used to construct the calibration model and twelve mixtures were used as a validation set. PCR and PLS multivariate calibration models were constructed and were found to have high spectral residuals in the regions below 210nm and above 350nm; so these regions were rejected.

For the selection of the optimum number of factors; the cross validation method was used leaving one sample out at each time, then predicting its concentration. The root mean square error of validation was calculated by comparing the predicted and the actual concentrations of each drug in the thirteen calibration mixtures. Data mean centering gave better results upon building PCR and PLS models.

2.4.2.2. Construction of the validation set

Twelve different mixtures containing different ratios of PC, DH, PAP and NOD were prepared and used as a validation set (table 2, mixtures 14-25); where the concentrations of all components were predicted using the developed models.

Table 1. Determination of paracetamol (PC) and diphenhydramine hydrochloride (DH) in presence of P-aminophenol (PAP) and the N-oxide degradation product of diphenhydramine (NOD) in laboratory prepared mixtures by the double divisor ²DD method.

Mixture No.	Ratio of PC:DH	Degradation products %	PC		DH	
			Taken (µgmL ⁻¹)	Recovery (%)	Taken (µgmL ⁻¹)	Recovery (%)
1	1:1	10%	36	99.583	36	99.417
2	2:1	20%	32	100.969	16	101.563
3	1:2	30%	7	100.857	14	100.357
4	5:1	40%	15	99.733	3	99.333
5	10:1	50%	20	99.750	2	99.500
6	20:1	60%	40	101.300	2	99.000
7	20:1	70%	40	99.650	2	101.500
8	20:1	80%	40	100.900	2	100.500
9	20:1	90%	40	99.325	2	101.000
Mean ± SD				100.230 ± 0.757		100.241 ± 0.973

Table 2. Mixtures of paracetamol (PC), diphenhydramine hydrochloride (DH), P-aminophenol (PAP) and N oxide degradation product of diphenhydramine (NOD) used in training and validation sets.

Mixture no.	Concentrations in µg mL ⁻¹			
	PC	DH	PAP	NOD
1	20	9	9	7
2	25	15	9	9
3	30	12	15	7
4	25	9	15	11
5	20	15	3	11
6	30	15	12	3
7	30	3	3	9
8	10	12	9	3
9	25	3	12	7
10	10	9	12	9
11	20	12	6	9
12	25	12	3	5
13	25	6	6	3
14	15	3	9	5
15	10	6	3	7
16	15	9	3	3
17	20	3	15	3
18	10	3	6	11
19	10	15	15	5
20	30	6	9	11
21	15	15	3	7
22	30	9	3	5
23	20	6	12	5
24	15	6	15	9
25	15	12	12	11

2.4.3. Application to pharmaceutical formulation; Panadol night[®] tablets

The contents of 10 Panadol night[®] tablets were powdered and mixed well. An accurately weighed portion of the powdered tablet equivalent to 100 mg of PC and 5 mg of DH was transferred into 100-mL

volumetric flask; 75 mL methanol was added and sonicated for 30 min, completed to volume with methanol, and then filtered. Part of the above solution is diluted to obtain working solution of the dosage form containing $100 \mu\text{g mL}^{-1}$ of PC and $5 \mu\text{g mL}^{-1}$ of DH. Applying the same procedure under double divisor ^2DD and multivariate spectral techniques, the concentrations of PC and DH were in the tablets were calculated from the computed regression equations.

III. Results and discussion

The main task of this work was to establish new simple, sensitive and accurate analytical methods for the simultaneous determination of PC and DH in presence of their degradates PAP and NOD in bulk powders and pharmaceutical dosage form with satisfactory precision for good analytical practice.

Upon adding 1 M NaOH to DH, the free base was liberated from the hydrochloride salt of DH. When 30% hydrogen peroxide solution is added to the free base and left for Six hours, complete oxidation took place into the N-oxide form (NOD) as shown in "Fig.2".

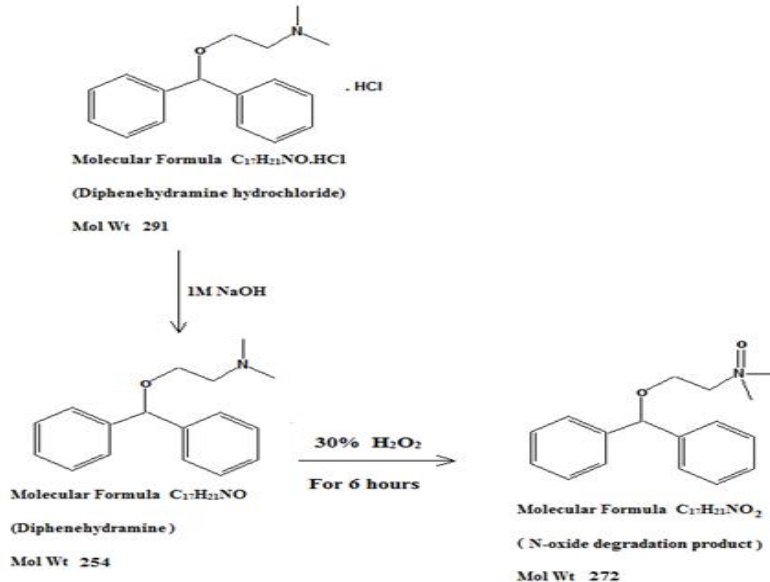
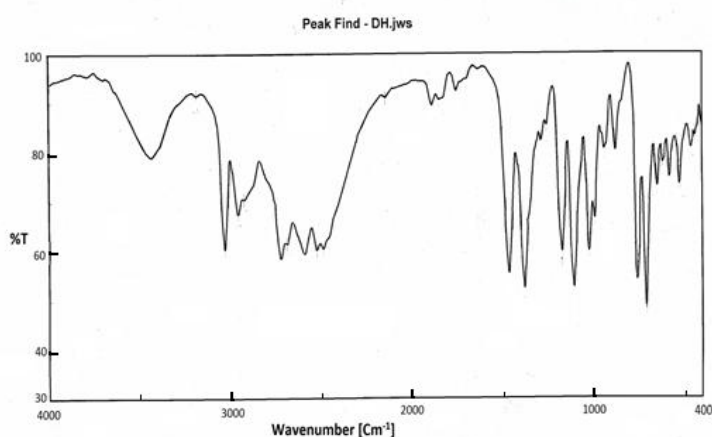


figure 2. scheme of oxidative degradation of diphenhydramine hydrochloride.

The solution was evaporated at room temperature and the degradation product (NOD) was identified by IR and mass spectrometry, where the appearance of a strong band corresponding to the N-oxide group at $1550/\text{cm}$ in the IR spectrum of NOD which is almost absent in the IR spectrum of DH is an evidence of the oxidation to the N-oxide form of the drug under the stated conditions as shown in "Fig.3".

The mass spectra of DH and NOD also confirmed their identity as mass molecular ion peaks at m/z 290 and 272 corresponded to the intact drug (DH) and its degradation product (NOD), respectively, as shown in "Fig.4".



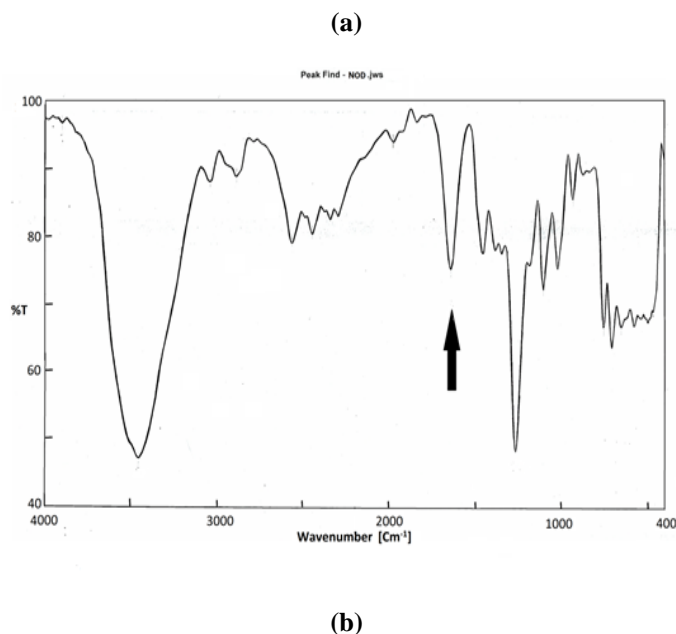


figure 3.infrared spectra of diphenhydramine hydrochloride (a) and its N-oxide degradation product Diphenylmethoxy -N,Ndimethylethanamine N-oxide (b).

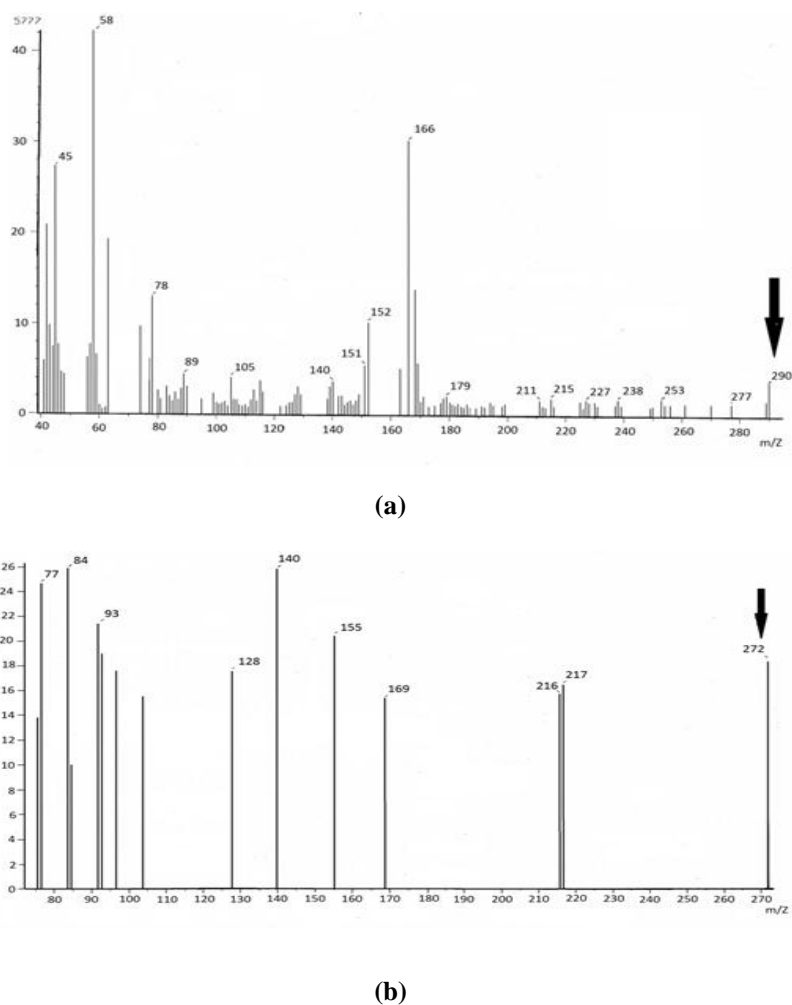


figure 4.mass spectra of diphenhydramine hydrochloride(a) and its N-oxide degradation product Diphenylmethoxy-N,N-dimethyl ethanamine N-oxide(b).

3.1. Double divisor ²DD spectral method

The double divisor derivative ratio spectral method depends on the use of coincident spectra of the derivative of the ratio spectra obtained by using a double divisor consisting of a sum of the spectra of two different components and measuring the amplitude at the maximum or the minimum wavelength.

The zero order absorption spectra of PC, DH, PAP and NOD show strong overlapping which prevents the direct spectrophotometric assay of PC or DH as shown in "Fig.5".

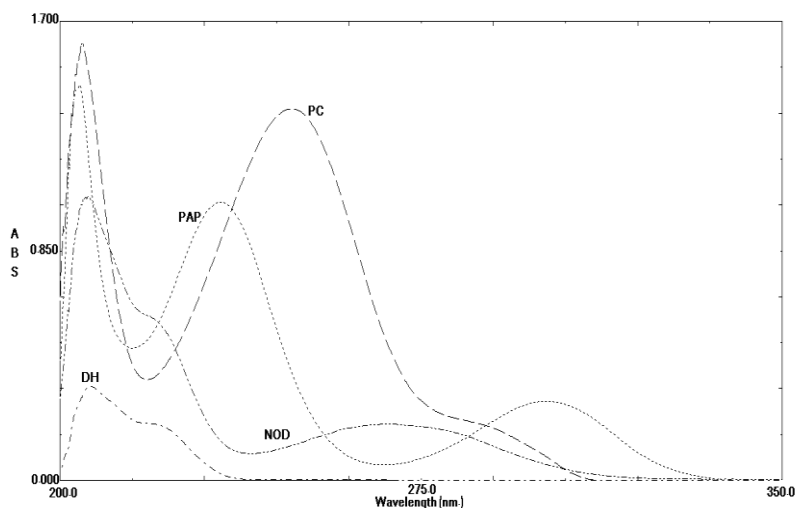


figure 5.the UV absorption spectra of paracetamol (- - -), diphenhydramine hydrochloride (....), p-aminophenol (.....) and N oxide degradation product of diphenhydramine hydrochloride (.....).

The double divisor derivative ratio technique was applied to solve this problem for selective determination of each of PC and DH in presence of the two degradates PAP and NOD.

Selection of the divisor: different concentrations of PAP and NOD mixtures were tried as divisors. The best results in terms of sensitivity, selectivity and signal to noise ratio were obtained upon using a mixture of PAP/NOD ($10\mu\text{g mL}^{-1}$ of each) as a divisor for determination of both PC and DH.

The absorption spectra were divided by the spectrum of a mixture of PAP/NOD ($10\mu\text{g mL}^{-1}$ of each) used as a divisor and ²DD spectra were obtained were PC could be determined at 304nm (a zero crossing point for DH) and DH could be determined at 256.4nm (a zero crossing point for PC) as shown in "Fig.6".

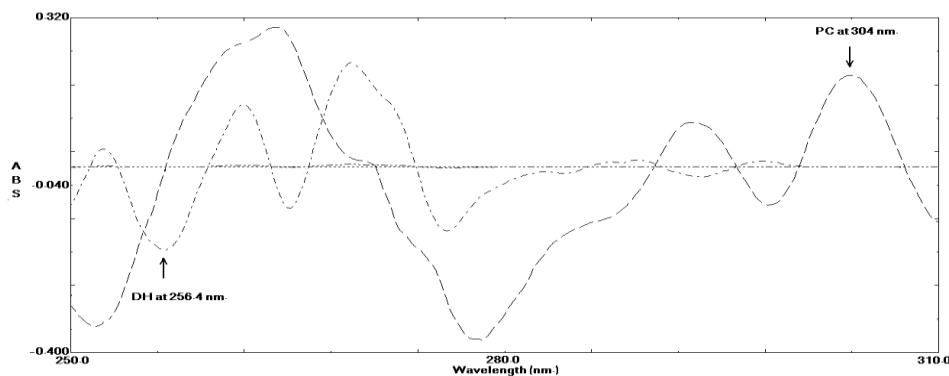


figure 6.the double divisor ²DD spectra of paracetamol (- - -), diphenhydramine hydrochloride (....), p-aminophenol (.....) and N oxide degradation product of diphenhydramine hydrochloride (.....), using a mixture of $10\mu\text{g mL}^{-1}$ of each of PAP and NOD as a divisor.

Linear correlations were found between concentrations of drugs and peak amplitudes at 304nm for PC and at 256.4nm for DH in the concentration range of ($2-45\mu\text{g mL}^{-1}$) for both drugs. The regression equations were calculated and found to be:

$$Y_1 = 0.0392 C_1 + 0.0021 (1) \quad r_1 = 0.9999 \quad \text{for PC and}$$

$$Y_2 = 0.0180 C_2 + 0.0038 (2) \quad r_2 = 0.9999 \quad \text{for DH}$$

where Y_1 and Y_2 are the peak amplitudes at 304 and 256.4nm, C_1 and C_2 are the concentrations in $\mu\text{g mL}^{-1}$, and r_1 and r_2 are the correlation coefficients for PC and DH, respectively.

Good linearity is evident from the high values of the correlation coefficients and low values of intercepts, “Table 3”.

The precision of the method was evident, as shown in “Table 3”. The methods accuracy was checked by applying the methods for determination of pure samples of the studied compounds. The concentrations were calculated from the corresponding regression equations and good recoveries are shown in “Table 3”.

Table 3. Results of assay validation parameters of the proposed double divisor ²DD method for determination of paracetamol (PC) and diphenhydramine hydrochloride (DH) in presence of P-aminophenol (PAP) and the N-oxide degradation product of diphenhydramine (NOD).

Parameter	PC	DH
Range (µg mL ⁻¹)	2-45	2-45
Linearity		
Slope	0.0393	0.0180
Intercept	0.0021	0.0038
Correlation coefficient	0.9999	0.9999
Accuracy (mean ± SD)	100.526±0.773	100.103±0.545
Precision (RSD%)		
Repeatability ^a	0.235	0.264
Intermediate precision ^b	0.275	0.436
^a The intraday precision (n = 3), average of three different concentrations repeated three times within one day.		
^b The interday precision (n = 3), average of three different concentrations repeated three times on three successive days.		

Accuracy was further assessed by applying the standard addition technique on Panadol ® tablets, for which good recoveries were obtained, revealing no interference from excipients, and good accuracy of the methods “Table 4”.

Table 4. Determination of paracetamol (PC) and diphenhydramine hydrochloride (DH) in Panadol night® tablets (batch no. 124992) by the proposed double divisor ²DD method and multivariate techniques and application of standard addition technique.

Components Taken µg mL ⁻¹	² DD method		PLS		PCR	
	Found (%±SD)	Standard addition mean±SD	Found (%±SD)	Standard addition mean±SD	Found (%±SD)	Standard addition mean±SD
PC 40	98.360 ±0.046	100.250 ±1.090	98.300 ±0.067	100.057 ± 1.875	98.320 ±0.067	100.833 ±1.258
DH 2	103.840 ±0.380	100.307 ± 0.992	103.750 ± 0.431	100.557 ±0.916	103.81 ±0.468	100.307 ±1.481

Selectivity of the proposed method is evident from the ²DD spectra in “Fig. 6”. Also, the method was applied for determination of PC and DH in laboratory-prepared mixtures as shown in “Table 1”.

3.2. Chemometric method:

Different chemometric techniques were used for the simultaneous determination of PC, DH, PAP and NOD including PLS and PCR. The first step of the determination is the construction of the calibration set which consists of 13 mixtures containing different ratios of PC, DH, PAP and NOD, as shown in “Table.2 ,mixtures 1-13”. The best results were obtained upon using the wavelength range of 210-350 nm.

A validation set including 12 mixtures was used to validate the predictability of the models, where good results were obtained “Table 2, mixtures 14-25”.

Predicted concentrations were plotted against actual concentrations to determine if the models accounted for the variation in the concentrations in the validation set.

The root mean square error of prediction [RMSEP] is a diagnostic tool which examines the errors in predicted concentrations, it indicates both precision and accuracy [10]. Model validation parameters are shown in “Table 5”.

Table 5. Results of assay validation parameters of the proposed multivariate methods for the determination of paracetamol (PC) and diphenhydramine hydrochloride (DH) in presence of P-aminophenol (PAP) and the N-oxide degradation product of diphenhydramine (NOD).

Validation parameters	PLS				PCR			
	PC	DH	PAP	NOD	PC	DH	PAP	NOD
Mean ± SD	100.047 ±1.357	100.003 ±0.822	99.954 ±1.314	99.679 ±1.211	99.964 ±1.068	99.914 ±0.947	99.751 ±1.271	99.86 ±1.206
RMSEP	0.1903	0.0548	0.1038	0.0676	0.1485	0.0587	0.0887	0.0903
Predicted versus actual concentration plot								
a-Slope	0.9955	0.9922	1.0037	1.0054	0.9969	0.9932	1.010	1.0116
b-Intercept	0.0769	0.0453	-0.0253	-0.0485	0.0409	0.0330	-0.0685	-0.0744
c-Correlation coefficient (r)	0.9996	0.9999	0.9998	0.9998	0.9998	0.9999	0.9999	0.9996

Table 6 shows statistical comparison of the results obtained by the proposed method and the official method, which is an HPLC method that uses C₁₈ column and a mobile phase consisting of phosphate buffer-acetonitrile in the ratio of 94:6 v/v [3] when applied to pure PC and DH. The calculated t and F values were smaller than the theoretical ones, indicating that there is no significant difference between the proposed method and the established method with respect to accuracy and precision.

Table 6. Statistical comparison of the results obtained by the proposed methods and the established method for determination of pure paracetamol (PC) and diphenhydramine hydrochloride (DH).

Items	² DD method		PLS		PCR		established method ^c	
	PC	DH	PC	DH	PC	DH	PC	DH
Mean	100.720	100.130	100.047	100.003	99.964	99.914	100.11	99.96
SD	0.493	0.618	1.357	0.822	1.068	0.947	0.868	0.743
N	10	10	12	12	12	12	6	6
Variance	0.243	0.382	1.841	0.676	1.141	0.897	0.753	0.552
Student's t-test	0.470 (2.228) ^a	0.703 (2.228) ^a	0.921 (2.179) ^a	0.922 (2.179) ^a	0.806 (2.179) ^a	0.901 (2.179) ^a		
F-value	3.099 (3.217) ^b	1.445 (3.217) ^b	2.445 (2.996) ^b	1.225 (2.996) ^b	1.515 (2.996) ^b	1.625 (2.996) ^b		
^a Figures in parentheses represent the corresponding tabulated values of t at P = 0.05. ^b Figures in parentheses represent the corresponding tabulated values of F at P = 0.05. ^c HPLC method, ^[3] .								

Table 7 shows statistical comparison of the results obtained by the proposed method and the official method [3] when applied to PC and DH in Panadol night® tablets. The calculated t and F values were smaller than the theoretical ones, indicating that there is no significant difference between the proposed method and the established method with respect to accuracy and precision.

Table 7. Statistical comparison of the results obtained by the proposed methods and the established method for determination of paracetamol (PC) and diphenhydramine hydrochloride (DH) in Panadol night® tablets(batch no. 124992).

Items	² DD method		PLS		PCR		established method ^c	
	PC	DH	PC	DH	PC	DH	PC	DH
Mean	98.36	103.84	98.30	103.75	98.32	103.81	98.35	103.80
SD	0.045	0.380	0.067	0.431	0.067	0.468	0.064	0.421
N	6	6	6	6	6	6	6	6
Variance	0.002	0.144	0.004	0.186	0.004	0.219	0.004	0.177
Student's t-test	0.92 (2.23) ^a	0.86 (2.23) ^a	0.20 (2.23) ^a	0.84 (2.23) ^a	0.44 (2.23) ^a	0.95 (2.23) ^a		
F-value	2.00 (4.28) ^b	1.29 (4.28) ^b	1.00 (4.28) ^b	1.06 (4.28) ^b	1.00 (4.28) ^b	1.22 (4.28) ^b		

^aFigures in parentheses represent the corresponding tabulated values of t at P = 0.05.
^bFigures in parentheses represent the corresponding tabulated values of F at P = 0.05.
^cHPLC method,^[3].

IV. Conclusion

The present work provides new sensitive, accurate and selective analytical techniques for the determination of PC and DH in the presence of their degradates PAP and NOD in bulk powder or in pharmaceutical formulation. Application of the proposed method to the analysis of PC and DH in laboratory-prepared mixtures and pharmaceutical formulation shows that neither the degradation products nor the excipients interfere with the determination, indicating that the proposed methods could be applied as stability-indicating methods for the determination of pure PC and DH and in the presence of the mentioned degradation products either in bulk powder or in pharmaceutical formulations.

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