

## Bioactive Principles in Two Polyherbal Traditional Anti-Diabetic Formulations

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**Abstract:** Bioactive principles in two polyherbal traditional anti diabetic formulations of different plants used in the treatment of diabetes mellitus mixed in different ratios were characterized using Infrared spectroscopy. Six medicinal plants with proven anti diabetic and related beneficial effects were selected for the preparation of two mixtures *Acanthus montanus*, *Asystasia gangetica*, *Emilia coccinea*, *Hibiscus rosa-sinesis*, *Momordica charantia* (Bitter melon), and *Venonia amygdalina*. Mixtures of the All- Six (AS) herbal leaves recorded these compounds 3-beta-acetoxy-5-etienic, acid dihydroxyacetone, acetobromo-alpha-D-galactose, dihydroxyacetone, ethylacetohydroxamate, P-tolyacetone, 4-aminoacetophenone, dihydroxyacetone, ethylacetohydroxamate, ethyl-4-chloro-2-cyanoacetoacetate while mixtures of All-four (AF) herbal leaves recorded ethylacetohydroxamate, ethyl-4,4,4-trichloroacetate, 4 – amino – acetophenone, ethylacetohydroxamate, p – tolyacetone, thiophene-2-acetonitrile, ethyl-4-chloro-2-cyanoacetoacetate, acetobromo-alpha-D-galactose, ethylacetohydroxamate, 4-aminoacetophenone and thiophene-2-acetonitrile. Functional groups such as those of the nitriles, benzene, acetals, cyano-compounds, amines, amides, substituted and conjugated ketones and aldehydes, alkaloids, phenyl groups, chlorocompounds, bromo sugar, glycosides, thiophene derivatives, amino substituted compounds and indoles were identified.

**Keywords:** Bioactive compounds, characterization, diabetes, polyherbs.

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### I. Introduction

Herbs are any Plants used for flavouring food, medicine and perfume. Herbal drugs have been used since the inception of human beings, and as a result are almost as old as life itself. Plants have formed the basis of sophisticated traditional medicine systems that have been in existence for thousands of years and continue to provide mankind with new remedies. In the world today, natural products and their derivatives represent more than 50% of the drugs clinically used (Ogbuji *et al.*, 2017). Medicinal plants have attracted increased attention since they can serve as an excellently pool for the discovery of new drugs by the use of their bioactive compounds (Ogbuji *et al.*, 2017). According to World Health Organization (WHO,1976), traditional medicine is defined as the sum total of all knowledge and practices, whether explicable or not, used in diagnosis, prevention and elimination of physical, mental or social imbalance using natural products (especially plant materials).

Bioactive compounds are compounds synthesized by plants that have the potential to be used by human for a variety of applications. They are extra nutritional constituents that typically occur in small quantities in foods. The bioactive compounds vary widely in chemical structure and functions and are grouped accordingly. Bioactive compounds also have actions in the body that promote good health, and they have been studied in the prevention of cancer, heart disease, diabetes and other diseases. (National Cancer Institute, 2014). Examples of these bioactive compounds include Alkaloids, Flavonoids, Saponins, Tannins, Indoles, Lignans, Phenols, etc. Phenolic compounds including their subcategory flavonoids are present in all plants and have been studied extensively in cereals, legumes, nuts, olive oil, vegetables etc. Many of these phenolic compounds have been studied to have antioxidant properties. (Olabinri *et al.*, 2010). When these bioactive compounds are introduced into the body, they bind to a particular biochemical targets, most notably to protein involved in signaling by hormones and neurotransmitters. This process is essentially, the basis for the effect of herbal medicine.

Herbal drugs are prescribed widely because of their effectiveness, less side effects, broad range of action and relatively low cost. However, the non trial drugs are usually not evaluated for purity and consistency of active compounds. They often contain contaminants and might show batch-to-batch variation. The exact mechanism of action, for example in lowering blood sugar is often not known. In addition, these herbs may not work well for everyone and their overall effects may vary in individuals, due to lack of standardization. Side



In the present study, six plants have been selected for the preparation of two mixtures. The herbs of interest were *Acanthus montanus*, *Asystasia gangetica*, *Emilia coccinea*, *Hibiscus rosa-sinensis*, *Momordica charantia* (Bitter melon), and *Vernonia amygdalina*.

## II. Materials And Method

### Collection and preparation of herbal samples

Fresh leaf samples of *Acanthus montanus*, *Asystasia gangatica*, *Hibiscus rosa-sinensis*, *Momordica charantia* and *Vernonia amygdalina* and *Emilia coccinea* were collected from Alaenyi Ogwa in Mbaitoli L.G.A Imo state, Nigeria. The leaves were sorted, washed and dried in an oven at 60<sup>0</sup>C. Each of the sample was pulverized using a high speed grinder separately. A formulation of the pulverized samples comprising of *Acanthus montanus*, *Asystasia gangetica*, *Emilia coccinea*, *Hibiscus rosa-sinensis*, *Momordica charantia* and *Vernonia amygdalina* in the ratio of 1:1(4g each) (AS) and also a formulation of *Acanthus montanus*, *Hibiscus rosa-sinensis*, *Momordica charantia*, *Vernonia amygdalina* in the ratio of 1:1 (4g each) (AF) were made and stored in an air tight containers separately.

### Sample preparation

A 0.5g portion of the formulation (AS) and (AF) were grounded in an agate mortar with 1g of specially purified potassium bromide (KBr) to fine powder to remove scattering effect from large crystals. This powdered mixture was then pressed in a mechanical press to form a translucent pellet through which the beam of spectrometer can pass through.

### Scanning process

The pellet was placed on the center of the sample plate and the pressure arm was swung over the sample and adjusted until it touched the sample. The scanning process for the sample began as the pressure arm touched it to generate spectrum with different wave numbers for functional groups of different organic or biological compounds.

These functional groups were correlated with different biological compounds present from the software library of the Pelkin Elmer spectrum BX 11 spectrometer.

### Spectral data analysis and compound identification

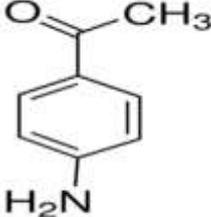
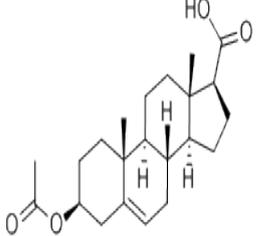
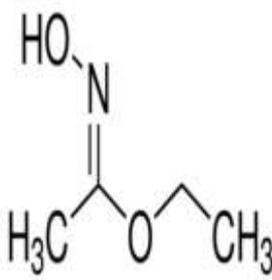
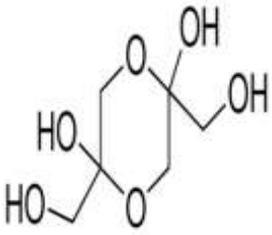
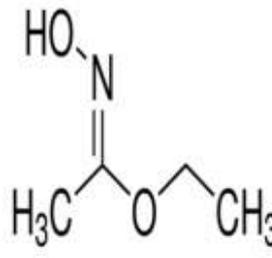
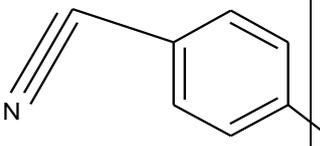
The spectra of the three samples obtained from the Pelkin Elmer spectrum BX 11 spectrometer was analyzed and compounds were identified by searching and matching the data on the NIST (National Institute of Standards and Technology) data base, Sigma-Aldrich online catalog product list, and pubchem compound search database. The results obtained from these searches were then used to characterize the bioactive compounds present in the samples.

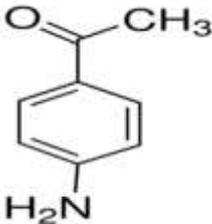
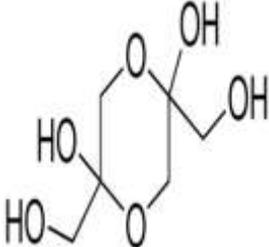
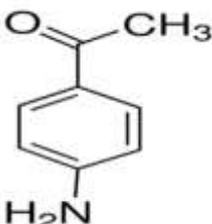
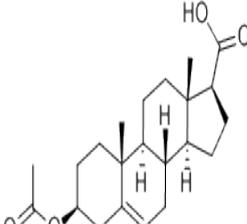
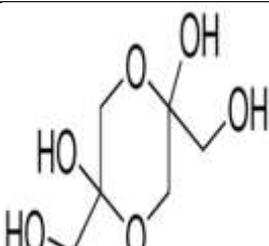
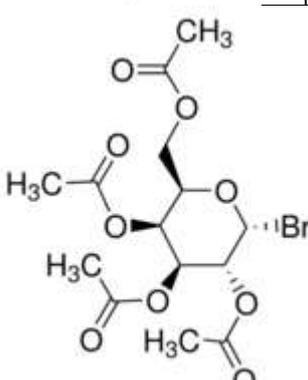
## III. Results

The spectra pattern and results of the spectral characterization of of the bioactive compounds in the All- Six (AS) herbal formulation are shown in fig. 1 and table 1 below.



Fig.1: Showing the Spectra Pattern of Some of the Bioactive Compounds in the All- Six (AS) herbal formulation.

SPECTRA WAVENUMBER (CM <sup>-1</sup> )	IDENTIFIABLE FUNCTIONAL GROUP	IDENTIFIED COMPOUND	STRUCTURE OF IDENTIFIED COMPOUNDS	MOLECULAR WEIGHT (g/mol)
1043.76	Primary amine	4-Aminoacetophenone		135.1632
1043.76	Ether	3-Beta-Acetoxy-5-Etienic Acid		360.491
1043.76	Ether.	Ethylacetohydroxamate		103.12
1043.76	Ether	Ethyl-4-Chloro-2-Cyanoacetate		189.59
1247.66	Ether	Dihydroxyacetone, Dimer		180.16
1324.61	Phenol or tertiary alcohol, OH bend	Ethylacetohydroxamate		103.12
1324.61	Aromatic tertiary amine	P-Tolylacetonitrile		131.1745

1324.61	Nitro group(primary amine)	4-Aminoacetophenone		135.1632
1643.36	Ether	Dihydroxyacetone, Dimer		180.16
1643.36	Ketone and Aldehyde	4-Aminoacetophenone		135.1632
1643.36	Ether	3-Beta-Acetoxy-5-Etienic Acid		360.491
2927.0	Alcohol/phenol	Dihydroxyacetone, Dimer		180.16
1643.36	Ester	Acetobromo-alpha-D-galactose		411.20

**Table 4.1:** showing characterized bioactive compounds from the All- Six (AS) herbal formulation.

Fig.2. The spectra pattern and results of the spectral characterization of some of the bioactive compounds in the All- Four (AF) herbal formulation.

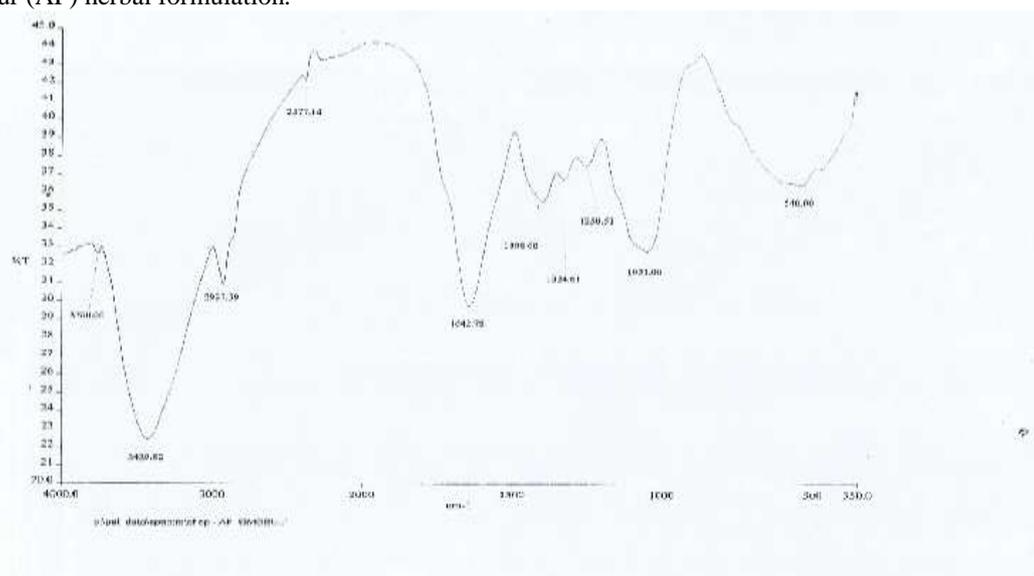
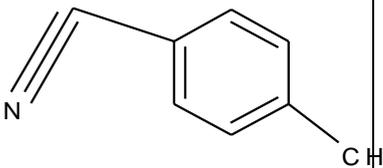
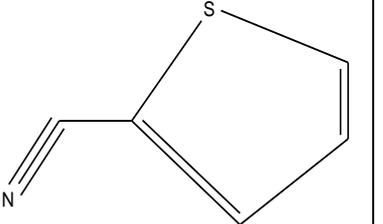
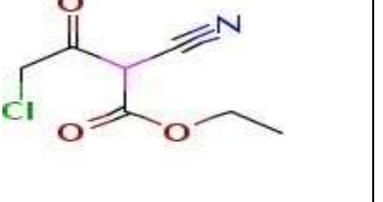
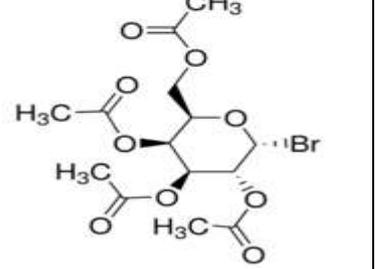
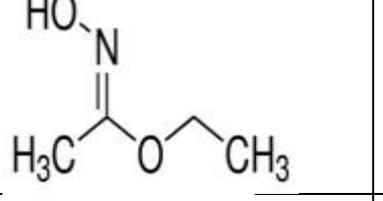
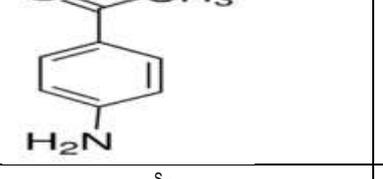
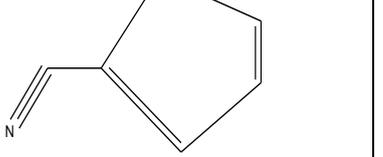


Fig. 2: Showing the Spectral Pattern of some of the Bioactive Compounds in the All-Four (AF) herbal formulation.

Table 2: Showing characterized bioactive compounds from the All- Four (AF) herbal formulation.

SPECTR A WAVE NUMBER (CM <sup>-1</sup> )	IDENTIFI ABLE FUNCTIO NAL GROUPS	IDENTIFIED COMPOUND	STRUCTURE OF IDENTIFIED COMPOUND	MOLECUL AR WEIGHT(g /mol)	BOND/VIBRATI ON
1051.0	Ethers	Ethylacetohydr oxamate		103.12	C – Br stretch C = O stretch C – O – C – O – C
1250.51	Ethers	Ethyl-4,4,4-Trichloroacetate		233.48	C – O stretch C – N stretch
1250.51	Aromatic primary amine	4 – Amino – Acetophenone		135.1632	C – H wag C – O stretch C = O stretch
1324.61	Phenol or tertiary alcohol, OH bend	Ethylacetohydr oxamate		103.12	C – H stretch C – H bend C - O stretch

1324.61	Aromatic tertiary amine	P Tolyacetonitrile		131.1745	N <sup>+</sup> H <sub>3</sub> deformation N – H bend -C = C stretch C = O stretch
1324.61	Aromatic tertiary amine	Thiophene-2-Acetonitrile		123.76	C ≡ N stretch C = C bend C – H stretch
1398.0	Aliphatic nitro compounds	Ethyl-4-Chloro-2-Cyanoacetoacetate		189.60	C ≡ N stretch C – S stretch C – H stretch
1642.78	Conjugated ketone	Acetobromo-alpha-D-galactose		411.20	O – H stretch C – O stretch
3429.82	Phenol, Alcohols	Ethylacetohydroxamate		103.12	
3429.82	Amide(primary)	4-Aminoacetophenone		135.1632	
1398.0	Nitro-group(nitrile)	Thiophene-2-Acetonitrile		123.76	

#### IV. Discussion

Figures 1 and 2 show the infrared spectra of the two mixtures (AS and AF). Table 1 (AS) shows these compounds; 3-beta-acetoxy-5-eticnic, acid dihydroxyacetone, acetobromo-alpha-D-galactose, dihydroxyacetone, ethylacetohydroxamate, P-tolyacetonitrile, 4-aminoacetophenone, dihydroxyacetone, ethylacetohydroxamate, ethyl-4-chloro-2-cyanoacetoacetate while table 2 (AF) shows the following compounds ethylacetohydroxamate, ethyl-4,4,4-trichloroacetate, 4 – amino – acetophenone, ethylacetohydroxamate, p – tolyacetonitrile, thiophene-2-acetonitrile, ethyl-4-chloro-2-cyanoacetoacetate, acetobromo-alpha-D-galactose, ethylacetohydroxamate, 4-aminoacetophenone and thiophene-2-acetonitrile. The compound p-tolyacetonitrile is

an acetonitrile in which the enzyme arylacetonitrilase catalyses its hydrolysis to arylacetic acid ammonia without any formation of amide. This enzyme does not attack the nitrile groups attached to aromatic and hetero-aromatic rings (Nagasawa. *et al.*, 1990). Arylacetic acids are useful intermediates in the synthesis of pharmaceuticals. P-tolylacetonitrile is involved in the synthesis of phenylacetic acid which is useful as an antiseptic (ELF, 1998). P-tolylacetonitrile has a boiling point of 242-243<sup>0</sup>C and melting point of 18<sup>0</sup>C and density of 0.992g/ml at 25<sup>0</sup>C. Acetobromo- $\alpha$ -D-galactose also known as Galactopyranosyl bromide is a versatile and important intermediate in carbohydrate chemistry (Gablíe and Deshmukh, 2010). It has been utilized as a starting material in the synthesis of thiogalactosides. Acetobromo- $\alpha$ -D-galactose and other related compounds are known as acetyl derivatives of carbohydrates. However, acetyl derivatives of carbohydrates are interestingly becoming important in medicinal chemistry and industries (Dandale. *et al.*, 2007). The compound was also identified in both mixtures.

Ethylacetohydroxamate is another compound which was identified in both mixtures, it belongs to the hydroxamic acid family. Hydroxamic acids are the family of compounds presenting the strong chelating properties towards various metal ions (Kurzak. *et al.*, 1992). Compounds containing the hydroxamic group have been shown to inhibit the activities of various metallo proteinases such as urease (Stemmler. *et al.*, 1995), oxidases (Ikeda-Saito. *et al.*, 1991), Alzheimer's Amyloid precursor protein secretase (Parvathy. *et al.*, 1998) or zinc proteinases involved in cancers (Hajduk. *et al.*, 1997; Groneberg. *et al.*, 1999). This compound has a melting point of 23-25<sup>0</sup>C, boiling point of 55-58<sup>0</sup>C and storage temperature of 2-8<sup>0</sup>C. It is moisture sensitive and a clear colourless liquid after boiling. The compound 4-aminoacetophene also known as p-aminoacetophenone has been reported to inhibit the growth of certain yeast strains including rad18 strain, bub3 strain, mec2-1 strain, rad50 strain in several researches of yeast anticancer drug screen (NCBI, 2014). According to a publication by the US national library of medicine (2013), p-aminoacetophenone is a powerful methemoglobin former in vivo. The compound is a slightly yellow to brown crystalline powder with a melting point, boiling point and density of 103-107<sup>0</sup>C, 293<sup>0</sup>C and 0.77g/cm<sup>3</sup> respectively.

2-Thiopheneacetonitrile is an organo-sulphur heterocyclic compound having a nitrile and sulphur group. It is a nitrile as well as a thiophene as it possesses the N $\equiv$ C group and one thiophene ring. Thiophene derivatives are found in natural plant pigments. Biotin, a water-soluble  $\beta$ -complex vitamin, is a reduced thiophene derivative. Thiophene moiety is found in cephalothin antibiotics. Thiopheneacetonitrile is used as an intermediate of antibiotics; cefoxitin, cephaloridine, and cephalothin ([www.Chemicaland2.com](http://www.Chemicaland2.com)). The compound is a clear colourless liquid with a boiling point of 115-120<sup>0</sup>C at 22mmHg and density of 1.157g/cm<sup>3</sup> at 25<sup>0</sup>C.

The compound 3-beta-acetoxy-5-etienic acid also known as 5-androstein-3-beta-ol-17-beta-carboxylic acid is an androstene ([www.CTD.com](http://www.CTD.com), 2014). Androstenes are unsaturated derivatives of the steroid androstane containing at least one double bond at any site in any of the rings. 3-beta-androstanediol as it is frequently called is an endogenous steroid hormone with an antiproliferative effect against prostate cancer (Weihua. *et al.*, 2002). It is a 5 $\alpha$ -reduced and 17 $\beta$ -hydroxylated metabolite of dihydroepiandrosterone (DHEA) as well as 3 $\beta$ -hydroxylated metabolite of dihydrotestosterone (DHT). Similarly to DHEA, 3 $\beta$ -diol is a high-affinity full agonist of the estrogen receptor 2 (ER $\beta$ ), and hence, an estrogen. In contrast, it does not bind to the androgen receptor (Oliveira *et al.*, 2007). This compound which was characterized in the sample AS has a melting point of 237-241<sup>0</sup>C and boiling point of 483.8<sup>0</sup>C at 760mmHg and a density of 1.16g/cm<sup>3</sup>. Dihydroxyacetone dimer is another compound characterize in the mixture AS. This compound is a simple carbohydrate, the normal form of Dihydroxyacetone (DHA) also known as glycerone. Its phosphate form dihydroxyacetone phosphate (DHAP) takes part in glycolysis, and it is an intermediate product of fructose metabolism ([www.wikipedia.com](http://www.wikipedia.com)). Recent studies have shown that DHA reacts with amino acids in the skin, which are part of the protein containing keratin layer on the surface. Various amino acids react differently with DHA, producing different tones of colouration (Faurshou and Wulf, 2004). The compound has a boiling point of 213.7<sup>0</sup>C at 760mmHg and melting point of 75-80<sup>0</sup>C. An acetoacetate called ethyl, 4,4,4-trichloroacetoacetate was characterized only in mixture AF. It is an ethyl ester of acetoacetic acid mainly used as a chemical intermediate in the production of compounds such as amino acids, analgesics, anti malarial agents, and vitamin B (Wilhelm and Hermann, 2005). It has a boiling point and density of 226.4<sup>0</sup>C and 1.437g/cm<sup>3</sup> respectively. The compound 2-Thenoyltrifluoroacetone as was characterized in mixture AF is an inhibitor of cellular respiration by blocking the respiratory chain at complex II. It binds at the quinone reduction site in complex II, preventing ubiquinone from binding ([www.wikipedia.com](http://www.wikipedia.com)). The compound is also used as a chelating agent pharmacologically.

## V. Conclusion

The characterization of the two mixtures of polyherbal antidiabetic formulation showed that there are sixteen bioactive compounds in both mixtures. Six of these secondary metabolites (bioactive compounds) are common in both formulations while four are not common in both formulations. Based on previous studies, these bioactive compounds characterized are of pharmacological and medicinal importance. A deeper insight into the isolation of the bioactive compounds characterized in these herbs may lead to the development of more potent anti-diabetic drugs.

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