Experimental Investigation on Domestic Refrigerator by Shell and Tube Heat Exchanger after the Condenser Using Sub Colling of Refregaring Fluid

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Abstract: Majority of refrigerator works on vapor compression refrigeration system. The system consists of compressor, condenser, expansion valve and evaporator. The performance of the system depends on the performance of all components of the system. The main objective of the present study is to study the performance of a domestic refrigerator by placing shell and tube type heat exchanger immediately after the condenser to extract more amount of heat by sub cooling process by using ammonia as an external cooling media.

Keywords: Sub cooling, Ammonia, Shell and tube heat exchanger, Coefficient of performance.

I. Introduction:
The term sub cooling refers to a liquid existing at a temperature below its normal saturation temperature. The sub cooling technology not only provides for additional capacity but also can reduce compressor power leading to higher overall system efficiency besides saving energy. The strategic placement of sub cooler for cooling in the liquid zone allow the operating pressure and temperature of the refrigerator system to be reduced and the refrigerator in the system to provide the highest cooling effect in the evaporator.

Fig. 1.1 vapor compression refrigeration system

Sub cooling desirable, reasons: It increases the efficiency of the system since the amount of heat being removed per kilogram of refrigerant circulated is greater. In other words we pump fewer refrigerants through the system to maintain the refrigerated temperature you want. This reduces the amount of time that the compressor must run to maintain the temperature.

Sub cooling is beneficial because it prevents the liquid refrigerant from changing to a gas before it gets to the Evaporator. Pressure drops in the liquid piping and the vertical risers can reduce refrigerant pressure to the point where it will boil or flash in the liquid line. The change of phase causes the refrigerant to absorb heat before it reaches the evaporator. Inadequate sub cooling prevents the expansion valve from properly metering liquid refrigerant in to the evaporator, resulting in poor system performance.

Selection of tube material: The tube material plays an important role in the design of refrigeration and air conditioning system. The tube material should have good thermal conductivity, as heat is transferred from hot to a cold side through the tubes, there is a temperature difference. Because of the tendency of the tube material the material is
thermally expands differently at various temperatures, and also thermal stresses will also occur during operation. All of these requirements call for careful selection of strong, thermally conductive, corrosion resistant, high quality tube material, typically metals including copper alloy, stainless steel, galvanized iron, carbon steel, nickel etc. Poor choice of tube material could result in a leak through the tube between shell and tube sides causing fluid cross-contamination and possibly loss of pressure.

![Shell and tube heat exchanger](image)

**Fig.1.2: Shell and tube heat exchanger**

**Shell and tube heat exchanger:**

A shell and tube heat exchanger is a class of heat exchanger designs. It is the most common type of heat exchanger and is suited for high pressure applications. As its name implies, this type of the heat exchanger consists of a shell with bundle of tubes inside it. One fluid runs through the tubes, another fluid flows over the tubes to transfer heat between the two fluids. The set of tubes is called tube bundle, and may be composed of Several types of tubes.

**Properties of ammonia:**

Molecular formula NH3 Molar mass 17.031 g/mol Appearance Colorless gas Odour strong pungent odour Density 0.86 kg/m3 (1.013 bar ) Melting point −77.73 °C Boiling point −33.34 °C Solubility in water 47% (0 °C) Solubility soluble in chloroform, ether, ethanol, methanol

<table>
<thead>
<tr>
<th>Structure</th>
<th></th>
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<tr>
<td>Solubility in water</td>
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<td>soluble in chloroform, ether, ethanol, methanol</td>
</tr>
</tbody>
</table>

**Thermo chemistry**

Flash point | flammable gas (see text) |
Explosive limits | 15–28% |
Ammonia, or azane, is a compound of nitrogen and hydrogen with the formulae NH₃. It is a colorless gas with a characteristic pungent smell. NH₃ boils at −33.34 °C (−28.012 °F) at a pressure of one atmosphere, so the liquid must be stored under pressure or at low temperature. Household ammonia or ammonium hydroxide is a solution of NH₃ in water. Measured in units of the Baumé scale (density), with 26 degrees baumé (about 30% (by weight) ammonia at 15.5 °C or 59.9 °F) being the typical high-concentration commercial product.

Solvent properties:
Ammonia is miscible with water. In an aqueous solution, it can be expelled by boiling. The aqueous solution of ammonia is basic. The maximum concentration of ammonia in water (a saturated solution) has a density of 0.880 g/cm³ and is often known as '.880 ammonia'. Ammonia does not burn readily or sustain combustion, except under narrow fuel-to-air mixtures of 15–25% air. One of the most characteristic properties of ammonia is its basicity.

II. Experimental Setup
The experimental test rig has a vapor refrigeration cycle working with refrigerant R-134a it consists of a compressor, evaporator, air cooled condenser, receiver, expansion valve and two shell and tube heat exchangers, One placed after condenser and the other after the evaporator. In the present work on the domestic refrigerator placing of shell and tube heat exchanger after the condenser contains ammonia that absorbs the heat from the refrigerating fluid coming from condenser. Sub cooling will occur and allow the operating pressure and temperature to be lowered. The net refrigeration effect and overall performance of the system will increase. The ammonia that is present in the modified shell and tube heat exchanger is converted from liquid to vapor state. The vaporized ammonia exchanges heat with the refrigerant coming out from the heat exchanger placed immediately after the evaporator is converted in to liquid .The cooled ammonia returns back to shell and tube heat exchanger after the condenser.

Modifications made to the existing system
Heat exchanger (HE-1) placed after the expansion valve in which liquid ammonia converts to vapor ammonia. Heat exchanger (HE-2) placed after the evaporator in which vapour ammonia converts to liquid. The ammonia continuously transferred from HE-1 to HE-2 and vice-versa.
In the above fig. 1.4 the liquid-vapor phase refrigerant flows through the tube and the Ammonia flows outside the tubes but inside the shell (the shell side). Heat is transferred from refrigerant to Ammonia through the tube walls, either from tube side to shell or vice versa.

### III. Tabular Column:

<table>
<thead>
<tr>
<th>Performance with</th>
<th>Suction pressure (bar)</th>
<th>Discharge pressure (bar)</th>
<th>Temperature of vapour refrigerant before shells (°C)</th>
<th>Inlet temperature of compressor (°C)</th>
<th>Outlet temperature of compressor (°C)</th>
<th>Condenser outlet temperature before shell (°C)</th>
<th>Condenser outlet temperature after shell (°C)</th>
<th>Evaporator temperature (°C)</th>
<th>Coefficient of performance</th>
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<tbody>
<tr>
<td>Base line</td>
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<td>19.1</td>
<td>78.2</td>
<td>-</td>
<td>-</td>
<td>42.9</td>
<td>2.6</td>
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<tr>
<td>Ammonia</td>
<td>0.6</td>
<td>11.0</td>
<td>18.1</td>
<td>25.0</td>
<td>69.5</td>
<td>42.0</td>
<td>39.5</td>
<td>1.7</td>
<td>6.11</td>
</tr>
</tbody>
</table>

Graph 1: Comparison of Heat Rejection Rate for Existing System and Proposed System

Graph 2: Comparison of net refrigeration effect for existing system and proposed system

Graph 3: Comparison of power input per tone of refrigeration for existing system and proposed system
Graph 4 Comparison of coefficient of performance for existing system and proposed system

IV. Result And Discussions:

By placing shell and tube heat exchanger after condenser the refrigerant is sub cooled by 2.5 °C when ammonia is used as coolant. Due to sub cooling the net refrigeration effect is increased and the C.O.P of the system is increased. The increase in coefficient of performance is 18.4% when ammonia is used as coolant. Coefficient of performance of refrigerator will be more with ammonia when compared with existing system

References:

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