

## Potential applications of various fungal and bacterial agents in decontamination of agricultural soils: An overview

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**Abstract:** This review aims to elaborate the potential applications of different bacterial and fungal species in decontamination of agricultural soils, which have been polluted with the application of continuous and higher doses of inorganic pesticides in the agricultural fields and this can be considered as a modern phytoremediation approach. Biodegradation is an eco friendly, cost effective, highly efficient approach and can be considered as a superior alternative to physical and chemical methods which are not only technically laborious and costly, also are not sufficient to completely degrade organic and inorganic toxins from the soil. Development of experimental conditions in which all congruent biological agents are applied concurrently may be a promising strategy to enhance biodegradation. Much work remains to be done in carrying out field studies based on laboratory-scale results/experiments using plant-associated endophytic and rhizospheric bacteria to degrade a wide range of toxic compounds of concern in environmental soil before commercially viable systems. Application of strong and effective natural phytotoxins as a potent pesticide is also highly recommended.

**Keywords:** Biodegradation, pesticides, fungi, bacteria, phytoremediation, phytodegradation.

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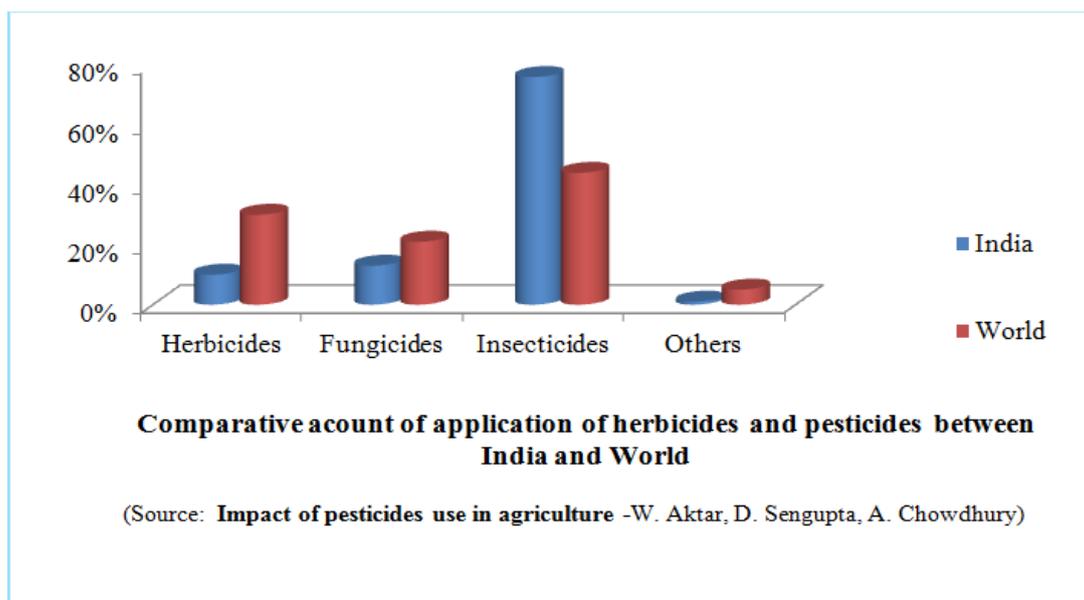
### I. Introduction

With the gradual development of civilization, man has realized the extent to which the pests harm his crops, annoy him and transmit diseases, both human and those of domestic animals. Neither governments nor individuals can tolerate the loss of crops which cost money, cause diseases and efforts to grow. The rapidly growing industrialization, increasing global population and higher demand of food leads to increase of food production through intensive agricultural practices, attention of public health and proper utilization natural resources. The improvement of agriculture with advanced agricultural technology to meet this demand, keeping soil in its productive quality plays a dominant role for much of today's productivity. Although wide-scale application of pesticides and herbicides is an essential part of augmenting crop yields; excessive use of these chemicals leads to the microbial imbalance, environmental pollution and health hazards. Huge amount of pesticides are applied annually in modern agriculture to enhance the production through controlling harmful effects caused by the target organisms including insects, fungi, bacteria, viruses as well as grasses grown (commonly known as weeds) in between the economical crops (Liu and Xiong, 2001). Thus, the random and widespread use of man-made, hazardous "xenobiotic" chemicals has led to a remarkable effort to implement new technologies to reduce or eliminate these contaminants from the environment.

Civilization has been combating insects and other pests throughout history. The previous history records many examples of how pests have had major impacts on humans. However, in the past 50 years the use of chemical control has increased tremendously (Day *et al.*, 1997). The modern era of chemical pest control started during World War II, when the much maligned DDT played a major role in the health and welfare of soldiers who used it to control body lice and mosquitoes which transmitted major illnesses. Further developments of pesticides followed. Owing to their low cost, ease of application and their effectiveness, pesticides have become the primary means of controlling pest. However, many pesticides are persistent organic pollutants which remain in an unmodified form in the environment for a prolonged time span. Persistent organic pollutants can also be defined as chemical substances that persist in the environment, bioaccumulate through the food web, and pose a risk of causing adverse effects to human health and the environment. As a consequence, increased use of these chemicals has caused considerable environmental pollution and human health problems (Hayes, 1986; Jit *et al.*, 2011; Vijgen *et al.*, 2011). Hence, contamination of soil, sediment, ground waters and surface waters by persistent organic pollutants is as an important environmental concern world-wide. The term pesticide covers a wide range of compounds including insecticides, fungicides, herbicides, rodenticides, molluscicides, nematocides, plant growth regulators and others. An ideal pesticide should have the ability to destroy target pest quickly and should be able to degrade non-toxic substances as quickly as possible but create no harm to the non-target species, including human being. But in reality, this is not the case, so the controversy of use and abuse of pesticides has surfaced. The science is very much clear that even low levels of exposure can harm human health, and children are particularly vulnerable. The ultimate "sink" of the pesticides applied in agricultural field and public health care is soil. Soil being the storehouse of multitudes of microbes, in quantity

and quality, receives the chemicals in various forms and acts as a scavenger of harmful substances. The abusive use of pesticides for pest control has been widely used in agriculture. However, the indiscriminate use of pesticides has inflicted serious harm and problems to humans as well as to the biodiversity (Gavrilescu, 2005; Hussain *et al.*, 2009). The injudicious uses of pesticides potentially damage the quality of soil which eventually creates damages in ecosystem balance. These ill-effects of pesticides necessitated the research for evolving protection system. Commonly used pollution treatment methods (e.g. land-filling, recycling, pyrolysis and incineration) for the remediation of contaminated sites have also had adverse effects on the environment, which can lead to the formation of toxic intermediates (Debarati *et al.*, 2005). Moreover, these methods are also very much costly and sometimes difficult to execute, especially in extensive agricultural areas. The main objective of modern pesticide research is to develop new chemicals which attack highly specific targets and disappear rapidly afterwards without accumulating in soil for long time period. One promising treatment method is to exploit the ability of microorganisms to remove pollutants from contaminated sites, an alternative treatment strategy that is effective, minimally hazardous, economical, versatile and environment-friendly, is the process known as bioremediation (Finley *et al.*, 2010).

The pesticides that enter in the soil environment do get degraded naturally, by a number of means. But it is needed that these pesticides should be degraded in an accelerated manner. The accelerated degradation of pesticides can be attained either by adapting the type of organisms to a particular pesticide and multiplying their number in the pesticide environment or creating such a condition in the environment that leads to enhanced degradation of the pesticides. However, the required details are not available for all the known pesticides and there is still scope for more work to explore some situation under which a pesticide affect or do not affect a particular system. For example, there has been very limited study on the fate of degradation products of pesticides and their toxic effects. A number of degradation pathways are incompletely worked out or just proposed; like the pathway of the TMTD degradation is not fully worked out. The enzymes responsible for degradation of pesticides have not been studied in detail. Thus it is of interest to develop methods for degrading and removing pesticides not only once they have completed their function in the soil plant system, but also when they accidentally enter soils and waters.



**Table 1:** Classification of pesticides

Classification of Pesticides	Types	Examples
<b>Based on chemical structure</b>		
<b>I. Main group</b>	i. Organochlorine	DDT, BHC, Aldrin, Endosulfun
	ii. Organophosphorus	Malathion, Parathion, Monocrotophos
	iii. Carbamates	Aldicarb, Carbafuran, Carbaryl
	iv. Pyrethroids	Aldrin, Cypermethrin
	v. Nionicotonides	Acetamiprid, Imidiclorprid
<b>I. Other groups</b>	i. Organotin compounds	Triphenyltin acetate,
	ii. Organomercurial compounds	Ethyl mercuric chloride, Phenyl mercuric bromide
	iii. Dithiomate fungicide	Mancozeb, Zineb, Maneb
	iv. Chlorifenoxy compounds	2,4-D, 2,4,5-T, DCPA, MCPP

	v. Benzimidazole compounds	Benomyl, Carbendiazim
	vi. Dipyridiliums	Paraquat, Diquat
	vii. Miscellaneous	Bromoxin, Simazine
<b>Other classification of pesticides</b>	i. Allephatic compounds	Malathion, Glyphosate, Maneb
	ii. Aromatic compounds	2,4-D, Diuron
	iii. Heterocyclic ring compounds	Nicotin, Captan, Atrazine
<b>Based on mode of action</b>	i. Nerve poison	Organanochlorin, organophosphate compounds and Carbamates
	ii. Anticoagulants	Warfarin
	iii. Juvenile hormones	Azadirachtin, Fenoxycarb
	iv. Antifeedents	Neem citrus derivatives limonoids & their synthetic derivatives
	v. Repellents	Neem oil, Citronella oil
<b>Based on pesticide action</b>	i. Stomach action	DDT, BHC, Metolachlor
	ii. Contact insecticide	Aldrin, Parathion
<b>Chemical classification of organochlorine pesticides</b>	i. Cyclo diene compounds	Aldrin, Endosulfan
	ii. Halogenated aromatic	DDT, Kelthane
	iii. Compounds	Chlorobengylate
	iv. Cycloparaffins	HCH, Lindane
	v. Chlorinated terpens	Polychlorcamphenes, Polychlorpinenes

**Table 2:** Different types of Pesticides, their targets and some important examples

Number	Pesticides	Targets	Examples
1	Algaecide or Algicide	Algae	Benzalkonium chloride, Bethoxazin, Copper sulfate, Cybutryne, Dichlone, Dichlorophen, Diuron, Fentin, Endothal hydrated lime, Isoproturon, Methabenzthiazuron, Nabam, Oxyfluorfen, Pentachlorophenyl, laurate, Quinoclamine, Quinonamid, Simazine, Terbutryn, Tiodonium
2	Acaricide or Miticide	Mites or Ticks	Permethrin, Ivermectin, Carbamate, Formamidine, Dicofol
3	Avicide	Birds	Strychnine, DRC-1339, Avitrol, Parathione, Chloralose
4	Bactericide or Bacteriacide	Bacteria	Disinfectants(hypochlorites, chloramines, sodium perborate, ethanol, phenols), Antiseptics, Antibiotics(penicillin, daptomycin)
5	Virucide	Virus	Cyanovirin-N, Griffithsin, Scytovirin, EP 0978289 A1 with iodine, NVC-422, Zonrox bleach, Interferon, Lysol
6	Fungicide or Mycocide	Fungus	Amphotericin B, Candicin, Nystatin, Rimocidin
7	Insecticide	Insects	DDT, Sarin, Pyrethroids
8	Herbicide	Weeds	2,4-D, Imidazolinones, Triazolopyrimidines, Pyrimidinylxybenzoates, Sulfonylamino carbonyl triazolinones
9	Herpeicide	Reptiles	
10	Larvicide	Insect larva	Methoprene, Temephos
11	Molluscicide	Snail and slugs	Metaldehyde, Metal salts, Methiocarb
12	Nematicide	Nematodes	Carbamate
13	Ovicide	Eggs	
14	Piscicide	Fish	Rotenone, Saponins, TFM, Niclosamide and Antimycin A (Fintrol)
15	Rodenticide	Rodents	Aluminium phosphide, Calcium phosphate, Zinc phosphate, Arsenic trioxide, Endrin
16	Biopesticide	Green pest control	Algae, Fungi, Bacteria, Higher plants, etc.

## Bioremediation

Bioremediation is the process by which living organisms degrade hazardous pollutants. Among different bioremediation methods, microbial metabolism is accepted as a safer and efficient tool for the removal of many pollutants. The relatively inexpensive technology of bioremediation for reclaiming chemically contaminated land has therefore been steadily gaining acceptance since 1980's. Different bioremediation approaches have been successfully applied for the removal of soils contaminated with a variety of xenobiotic compounds (Newcombe and Crowley, 1999; Top *et al.*, 1999; Cunningham and Philip, 2000; Juhasz *et al.*, 2000; Runes *et al.*, 2001; Manzano *et al.*, 2003). Bioremediation may be applied after excavation of polluted site material and transport to a controlled environment (*ex situ*) or, under relatively natural conditions in the field (*in situ*).

## Potential of Microbial Agents in Decontamination of Soil

Although pesticides are known to be degraded by some selected microorganisms, but the use of microorganisms for the detoxification of pollutants in the environment is a recently acknowledged feasibility. The potential problems involved with the use of microorganisms in the environment have been discussed in reference to oil clean up (Cobet *et al.*, 1973; Zobell, 1973). The use of microbial populations for the cleanup of

oil spills has been reported. Some oil degrading microbial preparations have been reported to be commercially available (Atlas and Bartha, 1973). Only in the past twenty years or so, have investigations shown that soil undesirably contaminated with a pesticide could possibly be decontaminated by inoculation with specifically adapted microorganisms. Several soil bacteria can degrade 2,4-D (2,4-dichloro-phenoxyacetic acid) accumulated in the soil by microbial detoxification procedures (Audus 1964; Loos 1975; Smith 1989). Several *Arthrobacter*, *Flavobacterium*, and *Pseudomonas* spp. have been isolated from soil (Loos 1975) and used in liquid media to determine the metabolic pathway of 2,4-D solubilization and ultimate detoxification (Audus 1964; Evans *et al.* 1971; Loos 1975). With at least some of the degrading bacteria, release of inorganic chloride indicates complete breakdown of the 2,4-D molecule (Loos 1975; Kelly *et al.* 1989). Audus (1950) perhaps, first of all demonstrated that bacteria adapted to utilize the herbicide 2,4-D could accelerate the disappearance of this compound from soil. In another experiment Macrae and Alexander (1965) made an attempt to protect the seeds of alfalfa from the herbicide by inoculating the heavy suspension of *Flavobacterium* sp. capable of utilizing this herbicide. Kearney *et al.* (1969) showed accelerated DDT disappearance from flooded soil inoculated with DDT acclimated *Enterobacter aerogenes*. Daughton and Hsieh (1977) reported accelerated degradation of parathion in flooded and non-flooded soils by inoculation with parathion acclimated bacterial culture. Shirkot and Gupta (1985) have reported accelerated degradation of TMTD in soil by inoculation with TMTD-utilizing bacteria. The TMTD degrading *Pseudomonas aeruginosa* was extremely effective in rapidly degrading 300 to 500 ppm of TMTD in different soils within 3 weeks under laboratory conditions. Most of these studies have been carried out in the laboratory or pot house using various approaches including immobilized cells or enzymes, fixed microbial films, soil columns, and conventional waste water processes. In the case of soil, inoculation with natural isolates or constructed strains and stimulation of indigenous microbial populations have been the major approaches to removal of pesticides. Sims *et al.* (1986) have compiled a comprehensive list of methods for use in surface soil decontamination. Fixed films of bacterial cells on a variety of supports have been found successful in removing a wide range of pesticides and structurally related substances from water and waste water. Bouwer and McCarty (1982) reported the removal of chlorinated benzenes and aliphatics by a granular activated carbon column (CAC) with biological activity over a 2 year period at an efficiency level of 95 to 98 per cent. Smith and Mortensen (1991) carried out the degradation of 2,4-D residues using soil bacterium, *Pseudomonas testosteroni* in a sprayer tank containing an aerated solution of simple mineral salts and 2,4-D amine formulations. The organism used the herbicide as a carbon source. Rozgaj (1994) has also reported that pesticide degradation is mainly by biological means involving microorganisms (bacteria and fungi) in a contaminated system.

The biological methods are advantageous to decontaminate areas that have been polluted by pesticides. These methods consider the thousands of microorganisms in the environment that in order to survive seek for alternatives to eliminate the pesticides that were sprayed. It was well established that microbes have the ability to transform and/or degrade a number of chemical agents which are causing severe damage in the environment. Realizing that fact, scientists have been exploring the microbial diversity, particularly of contaminated areas in search of organisms that can degrade a wide range of pollutants. Hence, biotransformation of organic contaminants in the natural environment has been extensively studied to understand microbial ecology, physiology and evolution due to their bioremediation potential (Mishra *et al.*, 2001). The biochemical and genetic basis of microbial degradation has received considerable attention. Several genes/enzymes, which provide microorganisms with the ability to degrade organopesticides, have been identified and characterized. Many native microorganisms develop complex and effective metabolic pathways that permit the biodegradation of toxic substances that are released into the environment. Microbial degradation of chemical compounds in the environment is an important route for the removal of these compounds. The biodegradation of these compounds, i.e., pesticides, is often complex and involves a series of biochemical reactions. Although many enzymes efficiently catalyze the biodegradation of pesticides, the full understanding of the biodegradation pathway often requires new investigations. Several pesticide biodegradation studies have shown only the total of degraded pesticide, but have not investigated in depth the new biotransformed products and their fate in the environment. On account of the grave risks synthetic pesticides pose to the organisms, there is an incessant search for pesticide safety and for the development of sustainable agriculture. The biological pesticides are based on natural compounds that effectively control the infestation of pests in agriculture. The advantage is that, contrary to synthetic pesticides, they are efficient and does not cause collateral damage (Fravel, 2005; Gerhardson, 2002; Raaijmakers *et al.*, 2002; Meleiro *et al.*, 2010).

Among different organochlorine (OC) insecticides, used successfully in controlling a number of diseases, such as malaria and typhus, were banned or restricted after the 1960s in most of the technologically advanced countries. The introduction of other synthetic insecticides – organophosphate (OP) insecticides in the 1960s, carbamates in 1970s and pyrethroids in 1980s and the introduction of herbicides and fungicides in the 1970s–1980s contributed greatly to pest control and agricultural output. Ideally a pesticide must be lethal to the targeted pests, but not to non-target species, including man. Unfortunately, this is not the case, so the

controversy of use and abuse of pesticides has surfaced. The rampant use of these chemicals, under the adage, “if little is good, a lot more will be better” has played havoc with human and other life forms.

### **Role of fungi**

Fungi degrade pesticides by introducing minor structural changes to the pesticides rendering it non toxic and are released to soil, where it is susceptible to further biodegradation by bacteria. Several fungi such as *Agrocybesemiorbicularis*, *Auriculariaauricula*, *Coriolusversicolor*, *Dichomitussqualens*, *Flammulinavelupites*, *Hypholomafasciculare*, *Pleurotustosreatus*, *Stereumhirsutum*, and *Avatha discolor* have shown their ability to degrade various pesticide groups like phenylamide, triazine, phenylurea, dicarboximide, chlorinated and organophosphorus compounds (Bending *et al.*, 2002). According to Quintero *et al.*, 2007 several classes of pesticides such as lindane, atrazine, diuron, terbuthylazine, metalaxyl, DDT, gamma-hexachlorocyclohexane (g-HCH), dieldrin, aldrin, heptachlor, chlordane, lindane, mirex, etc. have been degraded to different extent by white-rot fungi. It is also well established that fungi isolated from oil spill environment can reduce oil pollution (Das and Chandran, 2011).

Pure cultures of the lignin-degrading fungi *Phanerochaetechrysosporium* and *Trametes hirsutum* have been shown to degrade Lindane (Bumpuset *al.*, 1985; Kennedy *et al.*, 1990; Mouginet *al.*, 1997; Singh and Kuhad, 1999). The proposed mechanism for degradation was similar to that of lignin degradation, i.e. multiple non-specific oxidative reactions resulting from generation of carbon-centered free-radicals (Bumpuset *al.*, 1985). The ability of several white rot fungi (*Pleurotussajorcaju*, *Pleurotusflorida* and *Pleurotuseryngli*) to degrade Lindane was reported by Arisoy (1998).

### **Role of bacteria**

Most bacterial species which degrade pesticides belong to the genera *Flavobacterium*, *Arthrobacter*, *Azotobacter*, *Burkholderia* and *Pseudomonas*. The nature of degradation varies among species and the target compound. *Pseudomonas* sp. and *Klebsiellapneumoniae* have been shown to possess hydrolytic enzymes that are capable of breaking down s-triazine herbicides, such as atrazine. Similarly, a number of enzymes such as oxygenases, hydroxylases, hydrolases and isomerases present in *Pseudomonas* sp. and *Alcaligenessp.* have been shown to degrade herbicide 2,4-D (Mulbry and Kearney, 1991). In soil environment, due to the limiting environmental conditions and nature of the pesticides, most of the pesticides undergo partial degradation leading to the formation and accumulation of metabolites in the soil system. These metabolites are sometimes more toxic and less soluble than the parent compound which may inhibit the microbial population in the soil and further reduce the degradation of the pesticide resulting in the partial degradation of the pesticide. For example, degradation of endosulfan in soil by fungi and bacteria follows such a pathway via oxidation and hydrolysis leading to the formation of toxic endosulfan sulfate and less toxic endosulfandiol (Weir *et al.*, 2006). Best example for partial degradation is DDT, which undergoes the degradation to form metabolites like DDD and DDE that are toxic and more persistent than the parent compound (Foght *et al.*, 2001). Introducing microbial population which can degrade the pesticides completely and by optimizing the environmental condition can enhance the degradation of pesticides and its metabolites in the soil. Microbial degradation of most of the recalcitrant organic compounds is limited by the presence of anionic species in the compound. The anions like chloride, sulphate etc. are strongly bonded to the hydrocarbon ring which prevents the microbes from attacking the ring structure. This may be due to increased toxicity of anionic group (Julia *et al.*, 2001).

## **II. Conclusion**

About 30% of agricultural produce is lost due to pests. Hence, the use of pesticides has become indispensable in agriculture. The objective in modern pesticide research is to develop chemicals which attack highly specific targets and disappear rapidly afterwards. Since these pesticides do not disappear rapidly, they have the ability to alter the ecological balance by attacking the non target system in the soil environment. One of the foremost effect of pesticides is inhibition of microbial population (Tu, 1991; Martinez-Toledo *et al.*, 1992; Gonzalez- Lopez *et al.*, 1993) in soil which may lead to elimination, decrease or modification of soil biological processes such as *Nitrification* (Tu, 1989; Lodhiet *al.*, 1994), *Ammonification* (Schuster and Schroder, 1990; Lafrance *et al.*, 1992), *Nitrogen fixation* (Cemakova *et al.*, 1993; Sarafet *al.*, 1994) etc. which are essential for soil fertility. The biochemical and genetic basis of microbial degradation has received considerable attention. Several genes/enzymes, which provide microorganisms with the ability to degrade organopesticides, have been identified and characterized. Thus, microorganisms provide a potential wealth in biodegradation. The ability of these organisms to reduce the concentration of xenobiotics is directly linked to their long-term adaptation to environments where these compounds exist. Moreover, genetic engineering may be used to enhance the performance of such microorganisms that have the preferred properties, essential for biodegradation (Schroll *et al.*, 2004). Exploitation of the ability of microorganisms to remove pollutants from contaminated sites, an

alternative treatment strategy that will be very much effective, minimally hazardous, economical, versatile and environment-friendly and it can be treated as a smart practice of bioremediation (Finley *et al.*, 2010).

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